

Degrees of Freedom in Distillation

To perform a simulation of a distillation column, a set of specifications needs to be provided. To do this, one needs to understand the concept of degree of freedom. This is defined as

$$\text{Degree of Freedom} = \text{Number of unknowns} - \text{Number of equations} \quad (2-1)$$

Consider a column with N_T trays and N_C components, total condenser and total reboiler (Figure 2-1). In the case of crude, one can consider the components to be the pseudo-components that are usually generated.

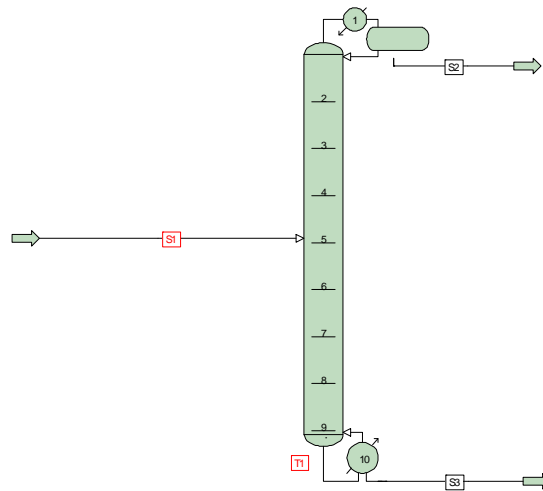


Figure 2-1: Single one-feed two product column with reboiler and condenser

VARIABLES

1) For each tray (excluding condenser and reboiler):

$$T_j, L_j, V_j, x_{j,i}, y_{j,i}$$

$$\text{Variables} = 3 N_T + 2 N_T N_C$$

L = Liquid rates

V = Vapor rates

x = liquid compositions

y = vapor compositions

T = Tray temperatures

2) For the condenser (temperature, flows, compositions and duty)

$$T_C, L_0, D, x_{C,i}, Q_C$$

$$\text{Variables} = N_C + 4$$

D = Product Rate

Q_C = Condenser heat duty

3) For the reboiler (temperature, flows, compositions and duty)

$T_R, V_{NT+1}, B, x_{Ri}, Q_R$

Variables = $N_C + 4$

B = Product Rate

Q_R = Condenser heat duty

Total number of Variables: $N_T(3+2N_C)+8+2N_C$

EQUATIONS:

1) Steady state mass balances for all components in all trays

$$L_{j-1} x_{j-1,i} - [L_j x_{j,i} + V_j y_{j,i}] + V_{j+1} y_{j+1,i} + F_j z_i = 0 \quad i=1, \dots, N_C \quad j=1, \dots, N_T$$

Equations: $N_C N_T$

z_i = feed composition

F_j = Feed rate to tray j (here we consider only one).

2) Steady state mass balances for all components the condenser

$$y_{1,i} = x_{Ci}$$

$$i=1, \dots, N_C$$

Equations: N_C

$$V_1 - (L_0 + D) = 0$$

Equations: 1

3) Steady state mass balances for all components the Reboiler

$$y_{NT+1,i} = x_{Ri}$$

$$i=1, \dots, N_C$$

Equations: N_C

$$L_{NT} - (V_{NT+1} + B) = 0$$

Equations: 1

4) Equilibrium Relations

$$y_{j,i} = K_{j,i}(x_j, T_j, P_j)x_{j,i}$$

$$i=1, \dots, N_C \quad j=1, \dots, N_T$$

Equations: $N_T N_C$

P_j = pressures (assumed given)

5) Summation equations

$$\sum_{i=1}^{N_C} x_{j,i} = 1 \quad j=1, \dots, N_T$$

$$\sum_{i=1}^{N_C} x_{C,i} = 1 \quad (\text{Condenser})$$

$$\sum_{i=1}^{N_C} y_{j,i} = 1 \quad j=1, \dots, N_T$$

$$\sum_{i=1}^{N_C} y_{NT+1,i} = 1 \quad (\text{Reboiler})$$

6) Enthalpy balance in each tray

$$L_{j-1} h_{j-1} - [L_j h_j + V_j H_j] + V_{j+1} H_{j+1} + F_j H_{F,j} = 0 \quad j=1, \dots, N_T$$

h : liquid enthalpy

H : Vapor enthalpy

H_F : Enthalpy of feed

All enthalpies are functions of composition and temperature.

7) Enthalpy balance in condenser and reboiler

$$V_1 H_1 - (L_0 + D) h_0 = Q_C \quad \text{Equations: } 1$$

$$(L_{NT} - B) h_{NT} - V_{NT+1} H_{NT+1} = Q_R \quad \text{Equations: } 1$$

Total number of equations: $N_T(3+2N_C)+2N_C+6$

Degree of Freedom: Unknowns - Equations = 2

Consider now the case of a column with a total condenser and steam injection, like the one in Figure 2-2.

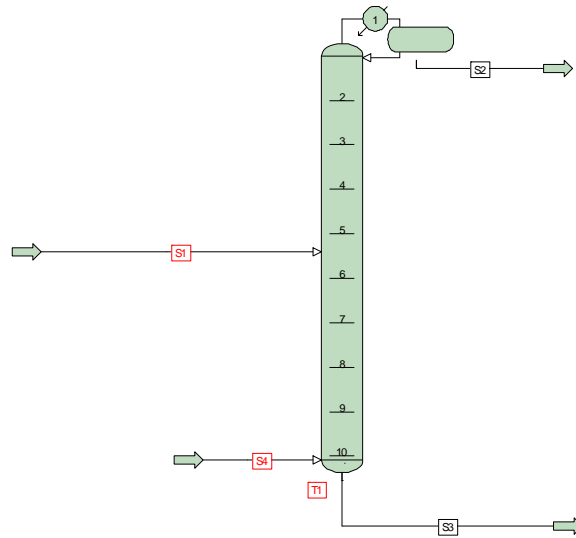


Figure 2-2: Single one-feed two product column with reboiler and condenser

Since there is no reboiler, there is N_C+4 variables less (T_R , V_{N+1} , x_{Ri} , B and Q_R). Therefore,

Total number of equations: $N_T(3+2N_C)+N_C+4$

The number of equations is reduced by N_C+3 (N_C+1 material balances, one summation of compositions, one energy balance). Thus,

Total number of equations: $N_T(3+2N_C)+N_C+3$

Degree of Freedom: Unknowns- Equations = 1