

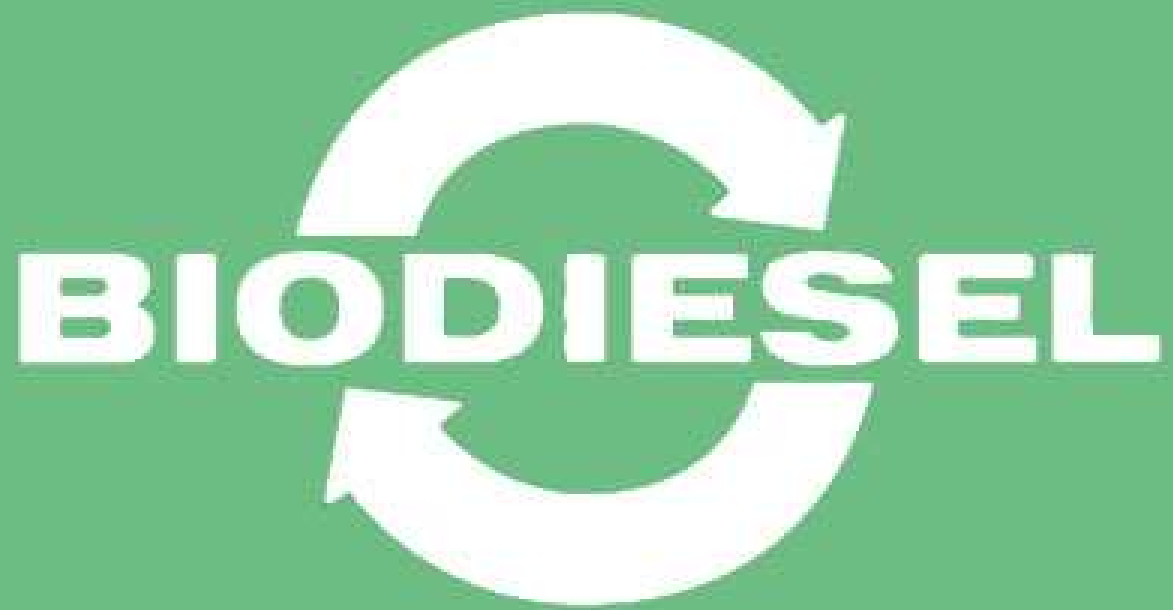
“And all I've outlined here today are interim strategies short-term and interim strategy. The truth of the matter is, the long-term strategy is to power our automobiles with something other than oil something other than gasoline, which is derived from oil.”

President Bush on Energy Policy

Marriott Wardman Park Hotel

Washington, D.C.

April 25, 2006



Grant Landrum

Joline Munoz

Outline

- Introduction to Biodiesel
- Biodiesel vs. Petrol Diesel
- Oil Sources
- Our Source
- Location Selection
- Genetic Modification
- Scenarios Evaluated
- Transesterification
- Byproducts
- Economics
- Conclusions

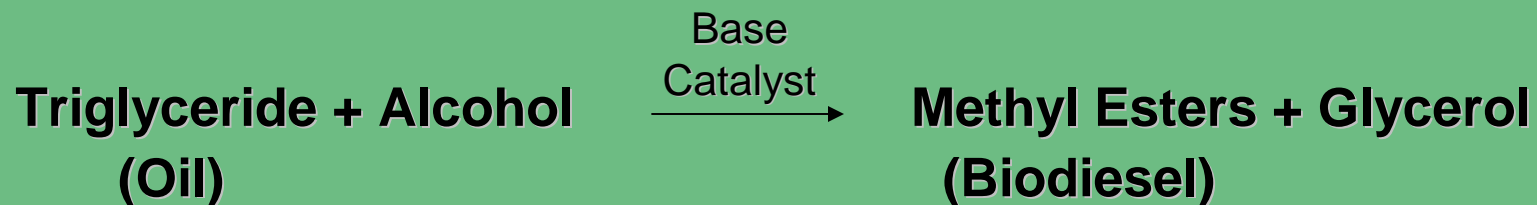


Introduction

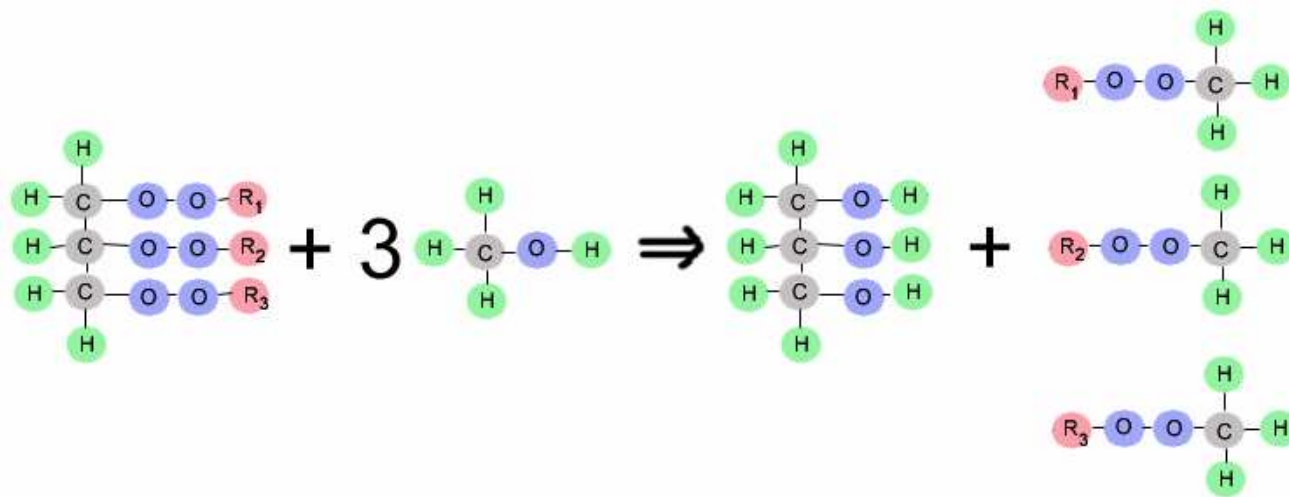
What is Biodiesel?

- Biodiesel is a completely natural, alternative fuel source that is applicable in almost all situations in which petrol diesel is used.

Transesterification



- *Oil can come from animal oil / fats, vegetable oil, and waste oil from restaurants (after degumming)*



Triglycerides

Alcohol

Glycerol

Methyl Esters

Biodiesel vs. Petrol-Diesel

- **Biodiesel**

- Reduces emissions
- Cetane #: 48-65
- Increases lubricity
- Longer engine life
- Higher Flash Point
- Requires no modification
- Dissolves rubber and some plastic
- Slightly lower heating value

- **Petroleum Diesel**

- High emissions
- Cetane #: 40-55
- Low sulfur diesel lacks lubricity
- High sulfur
 - poisons catalyst in exhaust system
 - Reduces effective reduction of emissions in exhaust system

Average Density and Heating Value of Biodiesel and Diesel Fuel

Fuel	Net Heating Value Avg., Btu/gal.	% Difference vs. No. 2 Diesel Avg.
No. 2 Diesel	129,500	
Biodiesel (B100)	118,296	8.65 %
B20 Blend (B20)	127,259*	1.73 %*
B2 Blend (B2)	129,276*	0.17 %*

* Calculated Values from those of No. 2 Diesel and Biodiesel (B100)

Cetane number = ease of ignition / smoothness of combustion.

High cetane = easy starting, at low temperature, low pressures, lower knocking characteristics.

Low cetane = poor ignition causes misfiring, rough operation, higher knocking.

**Acute Oral
Toxicity/Rates**

The lethal dose is greater than 17.4 g/Kg body weight.
table salt (NaCl) is 10 times more toxic.

**Skin Irritation -
Humans**

Very mild irritation. Less than 4 percent soap
and water solution.

Aquatic Toxicity

Deemed "insignificant" according to NIOSH (National
Institute for Occupational Safety and Health)

Biodegradability

Within 28 days, pure biodiesel degrades 85 to 88 %
in water.

Flash Point

Biodiesel's flash point is 260° Fahrenheit
Petroleum based diesel's is 125° Fahrenheit

Biodiesel Emissions		
(Shorter bars are better)		
Unburned Hydrocarbons	Diesel	Baseline
	B20	80%
	B100	8% to 33%
Carbon Monoxide	Diesel	Baseline
	B20	88%
	B100	35% to 52%
Particulate Matter	Diesel	Baseline
	B20	88%
	B100	42% to 53%
Sulfates	Diesel	Baseline
	B20	80%
	B100	10%
Ozone Potential of HC	Diesel	Baseline
	B20	90%
	B100	50% or less
NOx^a	Diesel	Baseline
	B20	102% (96%)*
	B100	110%

Potential Oil Sources



U.S. Department of Agriculture Statistics
 Canola: Area Planted, Harversted, Yield, and Production by State and United States, 2001-03

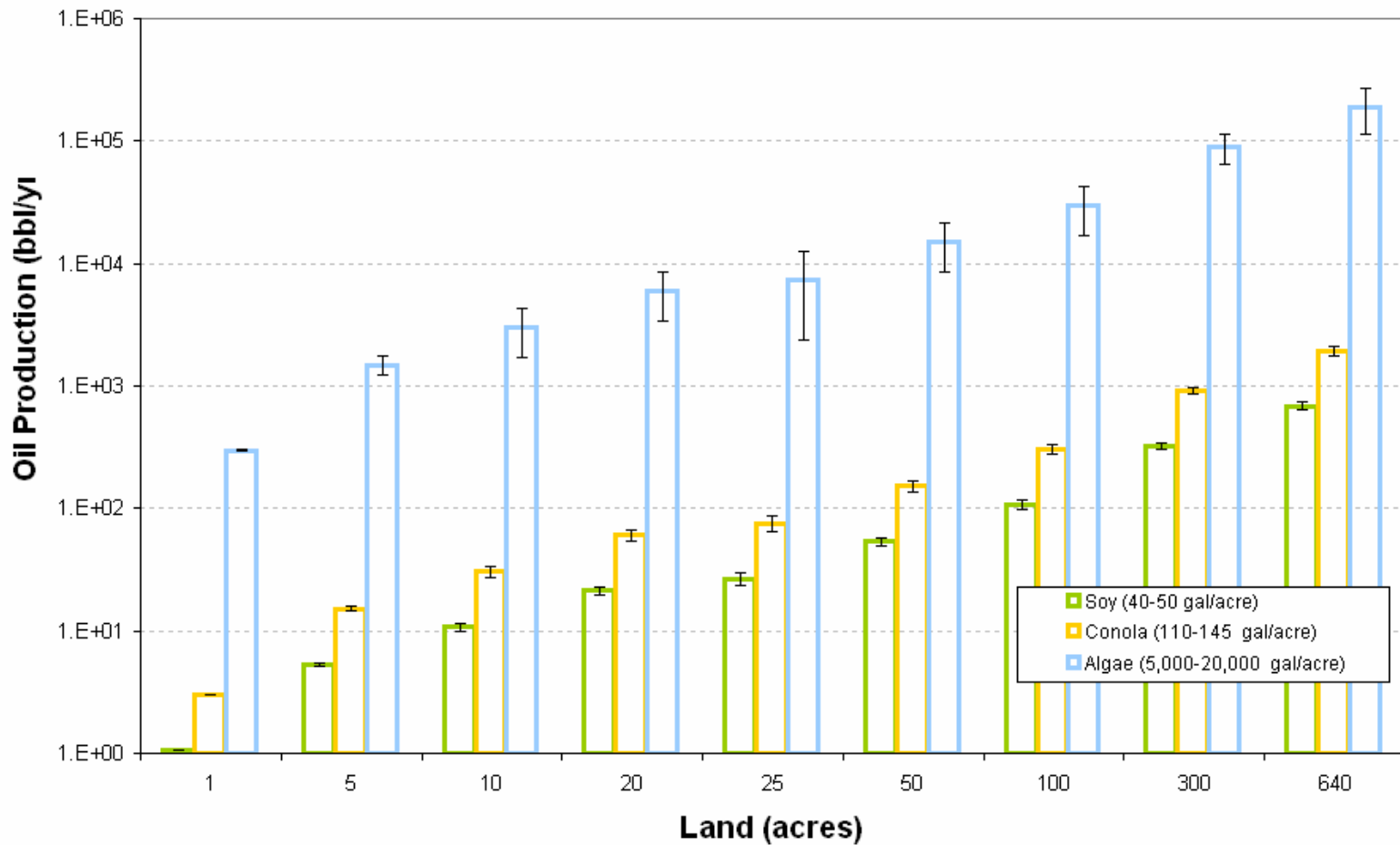
State	Area Planted			Area Harvested per 1,000 Acres		
	2001	2002	2003	2001	2002	2003
MN	80	80	57	75	45	56
ND	1,300	1,300	970	1,285	1,160	960
Other Sts 1/	114	79	55	95	70	52
US	1,494	1,459	1,082	1,455	1,275	1,068

Soybean

- Biodiesel plants in U.S. use soy
- Commodity Credit Corporation (CCC)
 - Incentives: \$1.45 - 1.47 / gallon till mid-2006
- 30% of the plants are not profitable without government incentives
- 24% are still unprofitable with incentives
 - High cost oil feed stock for biodiesel
 - High production costs
 - Increase lipid content-lower production costs

Land Requirements

Oil Production vs Land Required



Algae

- NREL Aquatic Species Program Study
(*National Renewable Energy Laboratory*)
 - Factors to consider
 - High adaptability to variety of environments
 - Large growth rate
 - Higher lipid content

Algae Requirements

- Growth $6\text{H}_2\text{O} + 6\text{CO}_2 \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2$
 - Sunlight
 - CO_2
 - Nutrients (N and Si)
 - Water (*can be saline*)
 - Warm Climate
 - Circulation or Aeration

Algae

- Environment
 - Different Media
 - Type 1: 2,000 to 16,000 mg/l
 - Type 2: 1,500 to 26,000 mg/l
 - Seawater
 - Majority algae survive well in seawater

Brine Water Source

- Hueco Bolson Aquifer

- Salt concentrations range from 1000 to 3000 mg/L

- 600 times the amount of fresh water
 - Possible alternative to evaporation ponds-reinjection

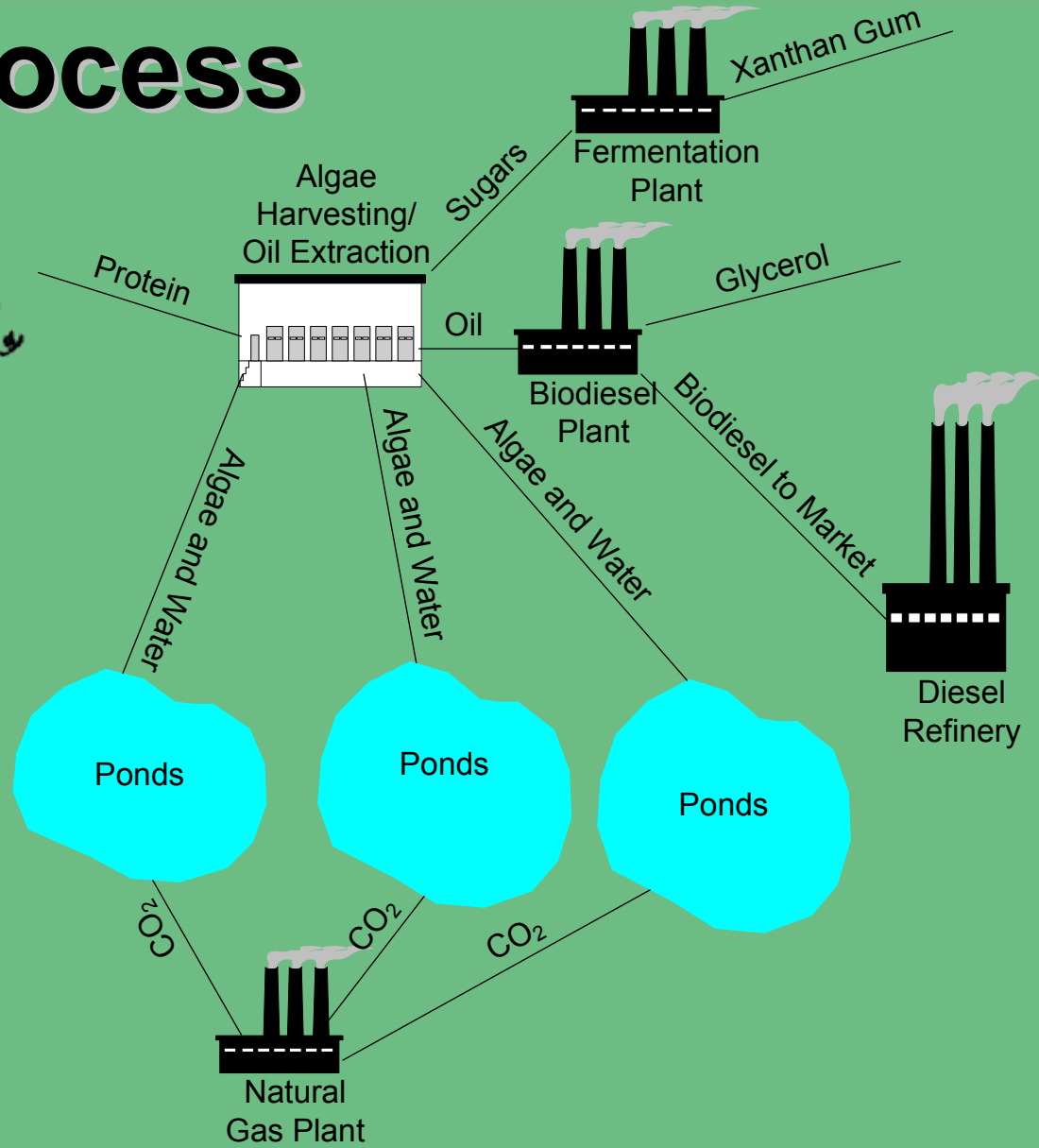
Hueco-Mesilla Bolson



Algae Selection

- Arizona State University - New Mexico / Texas Area
 - Amphora Bacillariophyceae
 - Hard to harvest (10 X 4 μm)
 - Salinity: ~9000-18000 mg/L
 - Oscillatoria Cynaophyceae
 - Adaptive
 - Harder to harvest (0.2 X 0.1 μm)
 - Salinity: ~2000-12000 mg/L

The Process



Location: West TX

- 24 oil refineries
- 8% diesel transportation



Newman, TX		
Stack	CO ₂	SO ₂
	<i>tons / year</i>	<i>tons / year</i>

1	90,069	0.4
2	88,858	0.5
3	109,273	0.6
4	130,436	0.7
5	135,758	0.6
SUM	554,395	2.8

Tolk Station, TX		
Stack	CO ₂	SO ₂
	<i>tons / year</i>	<i>tons / year</i>

1	1,537,026	4,543
2	2,112,768	5,993
SUM	3,649,794	10,536

Jones, TX		
Stack	CO ₂	SO ₂
	<i>tons / year</i>	<i>tons / year</i>

1	355,086	1.8
2	374,686	1.9
SUM	729,773	3.7

Newman Power Plant, El Paso TX.

Pond Nutrients

	%carbon	C _{algae} (lb/day)	%absorbed
1 mile ²	40%	80,000	3.9%
	50%	100,000	4.9%
	60%	120,000	5.8%

CO ₂ Available (lb/day)
7,515,059

C Available (lb/day)
2,051,611

4 mile ²	40%	320,000	15.6%
	50%	390,000	19.0%
	60%	470,000	22.9%

7 mile ²	40%	550,000	26.8%
	50%	690,000	33.6%
	60%	830,000	40.5%

The CO₂ utilization efficiencies of 96 ± 11% were achieved by bubbling CO₂ into the culture with the use of a counterflow sump system.

Concentration		
1 mi ²	4 mi ²	7 mi ²
M	M	M

SO ₂	2.71E-08	6.77E-09	3.87E-09
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SO ₂ Available (mol/day)
278

NO _x	1.00E-05	2.51E-06	1.44E-06
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NO _x Available (mol/day)
103,152

A Study of the Energetics and Economics of Microalgal Mass Culture with the Marine Chlorophyte *Tetraselmis suecica*: Implications for Use of Power Plant Stack Gases*

Algae

- **Lipid Content**

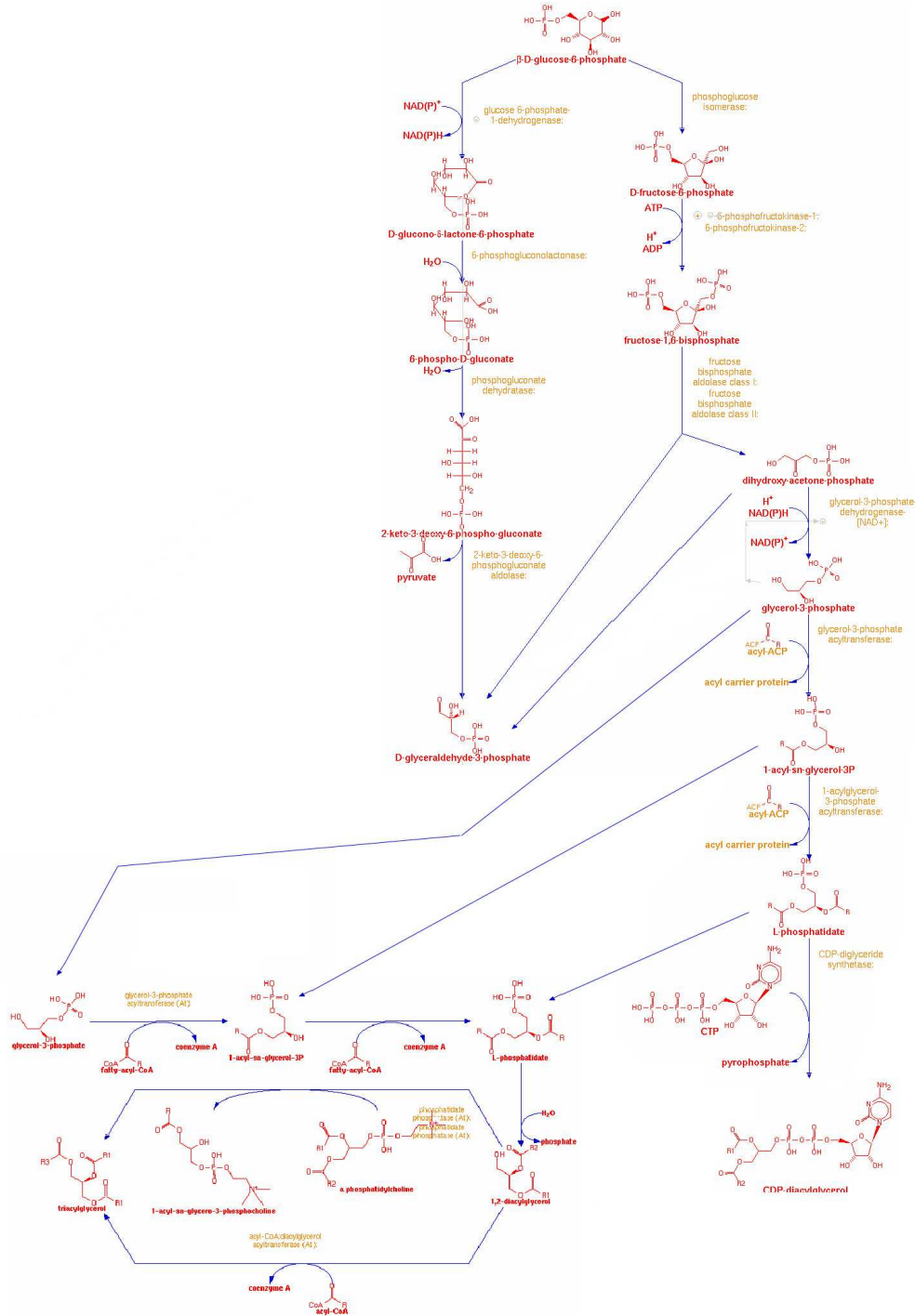
- **Starvation**

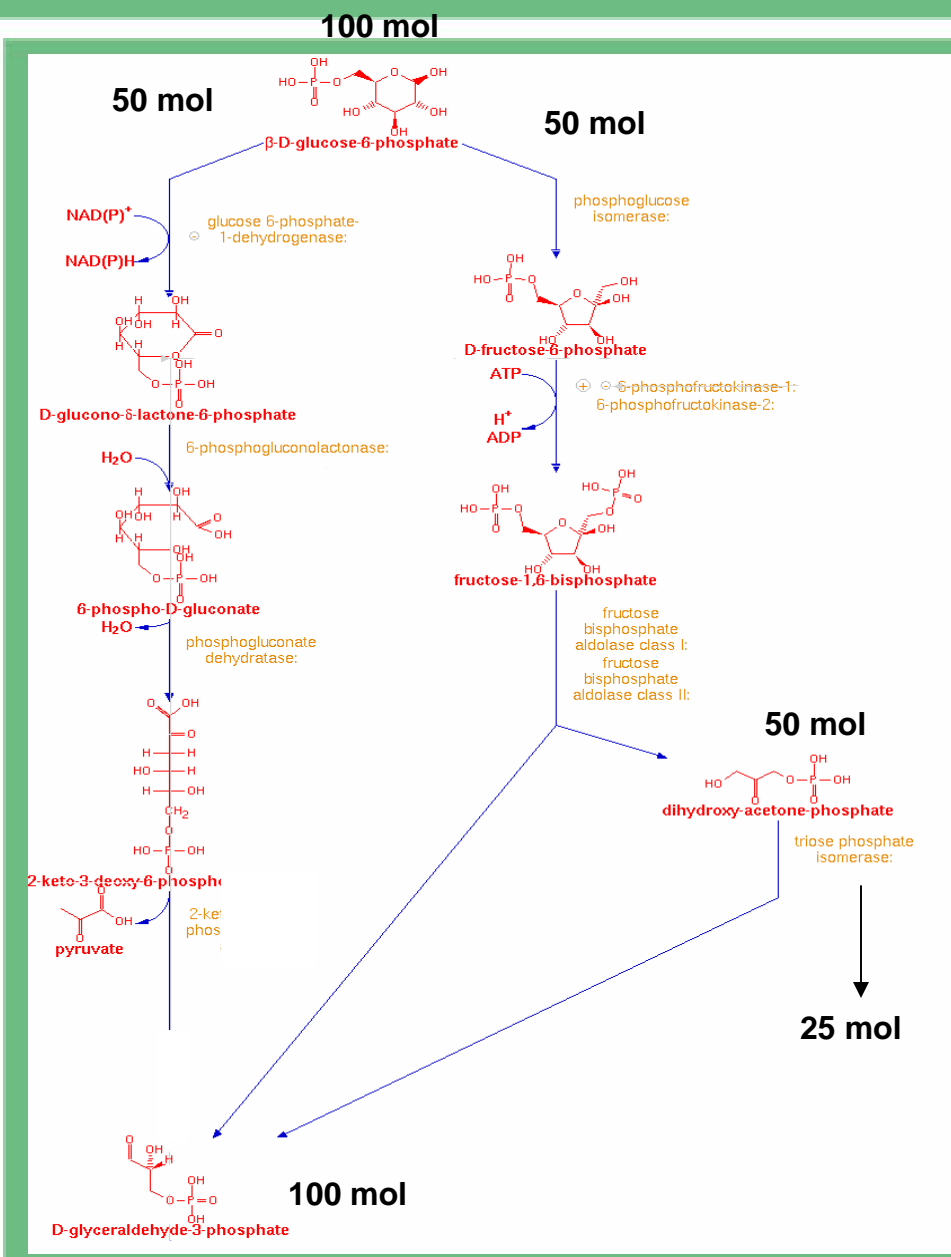
- A minimum amount of food (nitrogen source or Si source) added to ponds
 - 7-day cycle, algae have run out of food and have begun to starve
 - Pro-Causes many different algae to increase lipid content
 - Con-Stuns cell growth

- **Genetic Engineering**

- Metabolic processes in the specific algae
 - Manipulate these processes to maximize lipids

Overall Metabolic Pathway of Lipid Synthesis





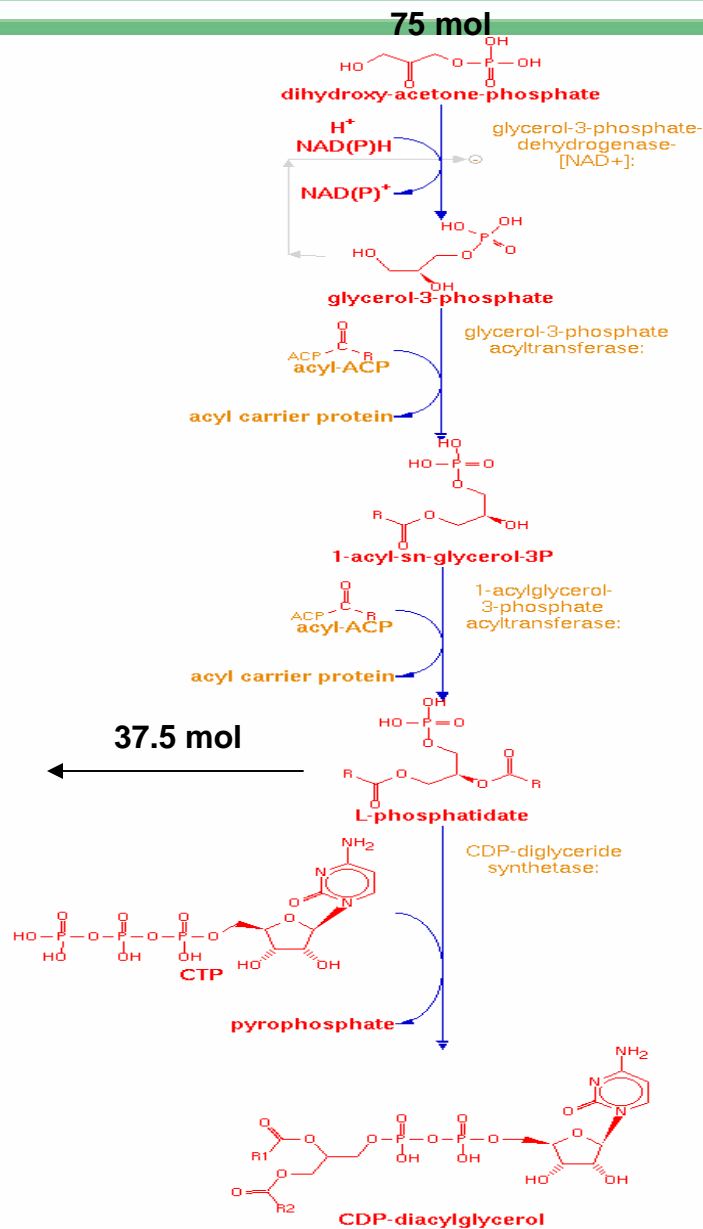
Entner-Doudoroff & Glycolysis I

Genetic engineering algae starting with glucose-6-phosphate

Want to produce dihydroxy-acetone-phosphate

Phospholipid Biosynthesis

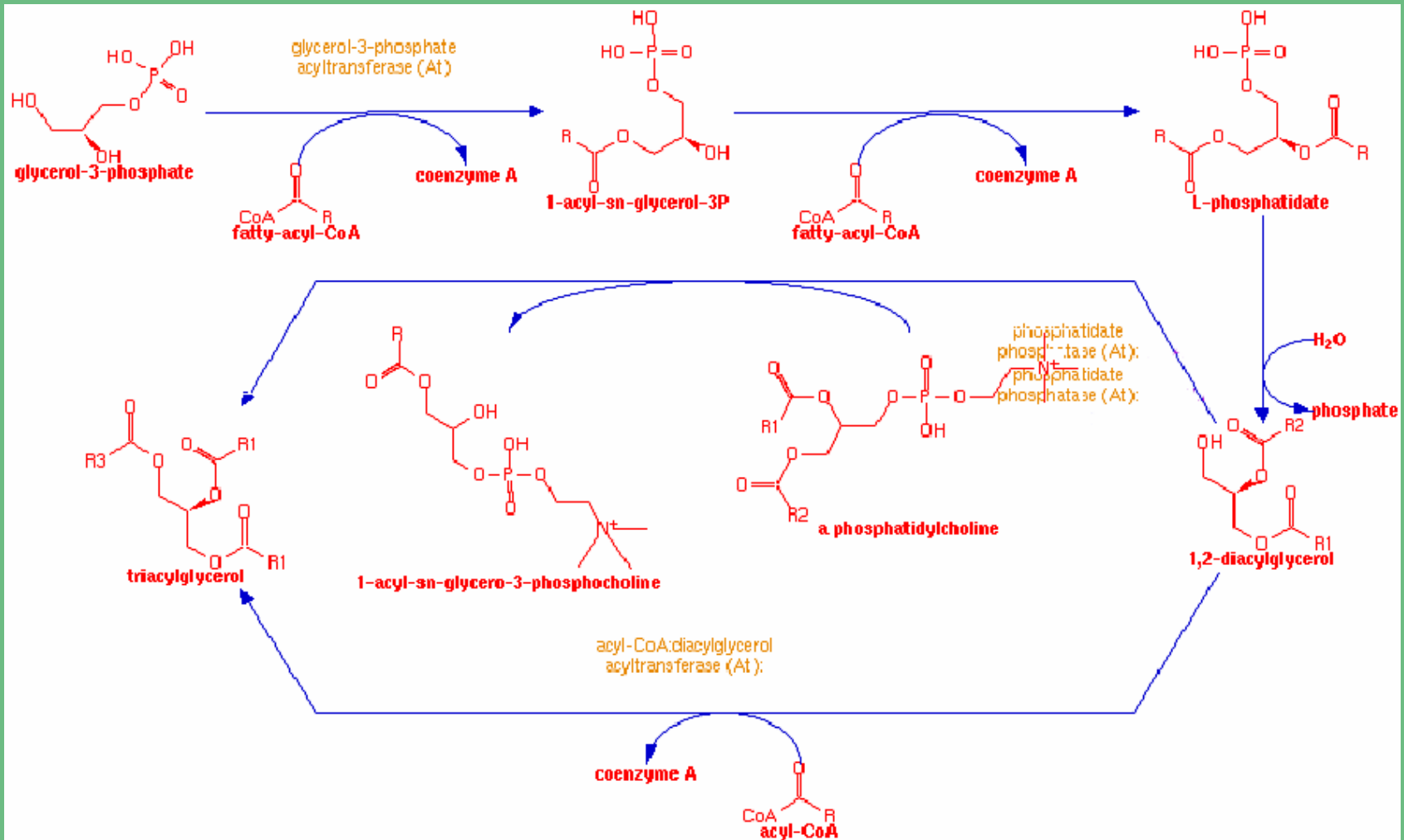
Want to produce
glycerol-3-phosphate
or L-phosphatidate



37.5 mol

Triacylglycerol Biosynthesis

Initially 37.5 mol tri / 100 mol G-6-P Now 56.25 mol tri / 100 mol G-6-P

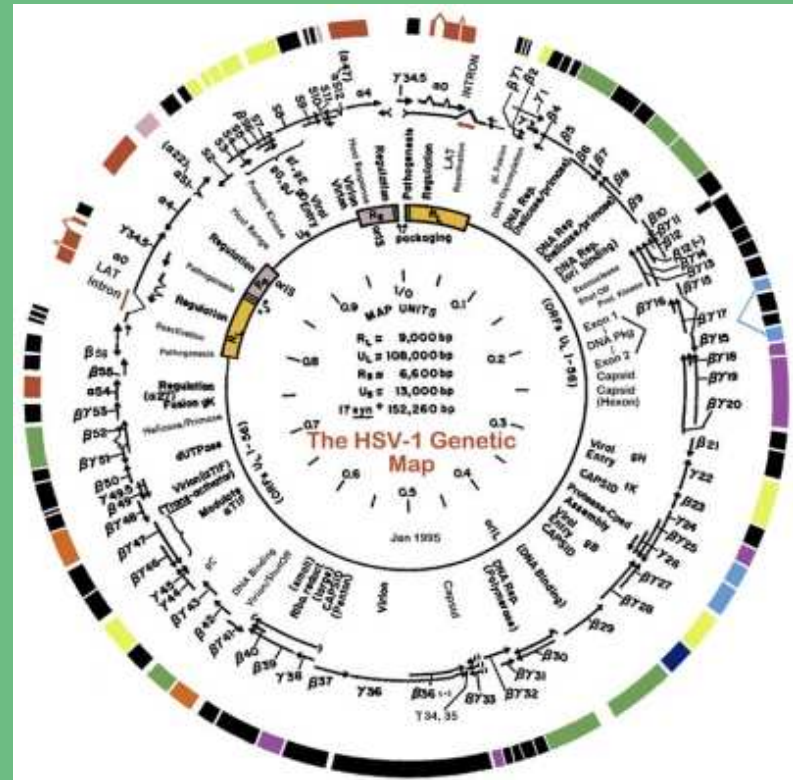


Genetic Engineering

- Why genetically modify the algae?
 - *Maximize algal lipid content*
- How to genetic engineer the algae?
 - *Complete genome study on the species*
 - *Isolation, Manipulation, and Re-introduction*
 - *Polymerase Chain Reaction (PCR)*
 - *Loss of Function / Gain of Function / Tracking*

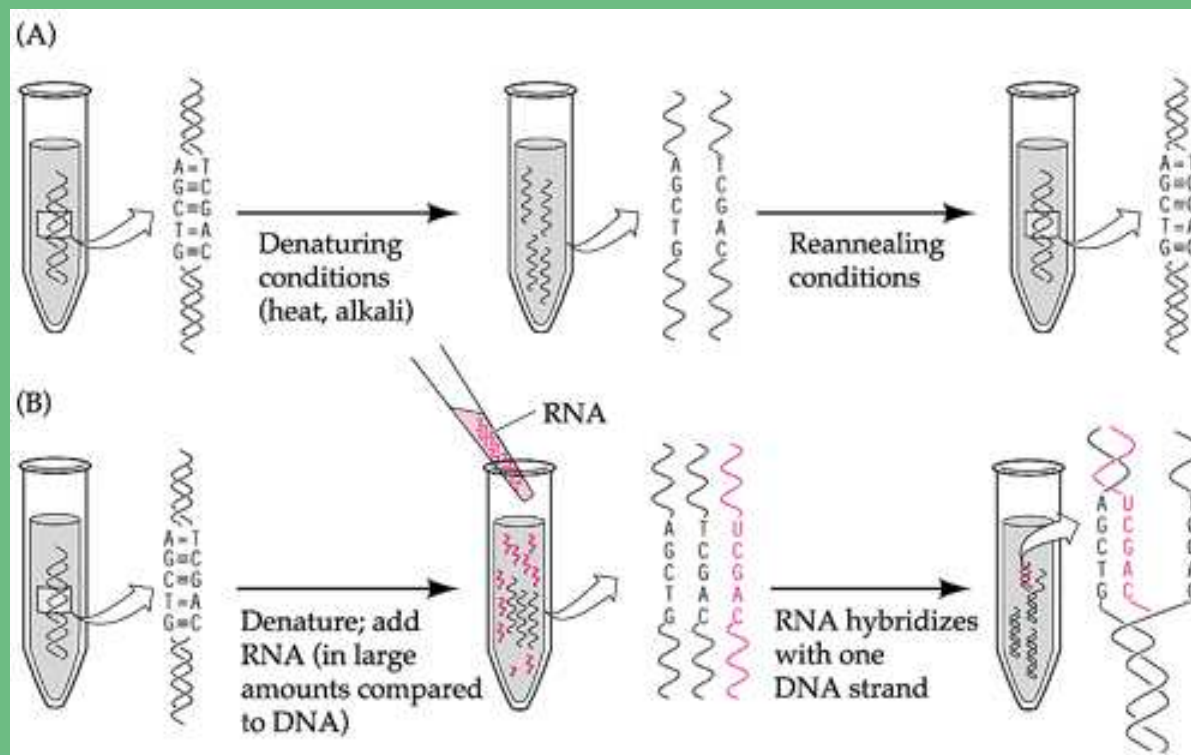
What is a genome?

- Contains all the biological information needed to build / maintain a living example of that organism
- Biological information in a genome is encoded in its DNA and is divided into discrete units called genes



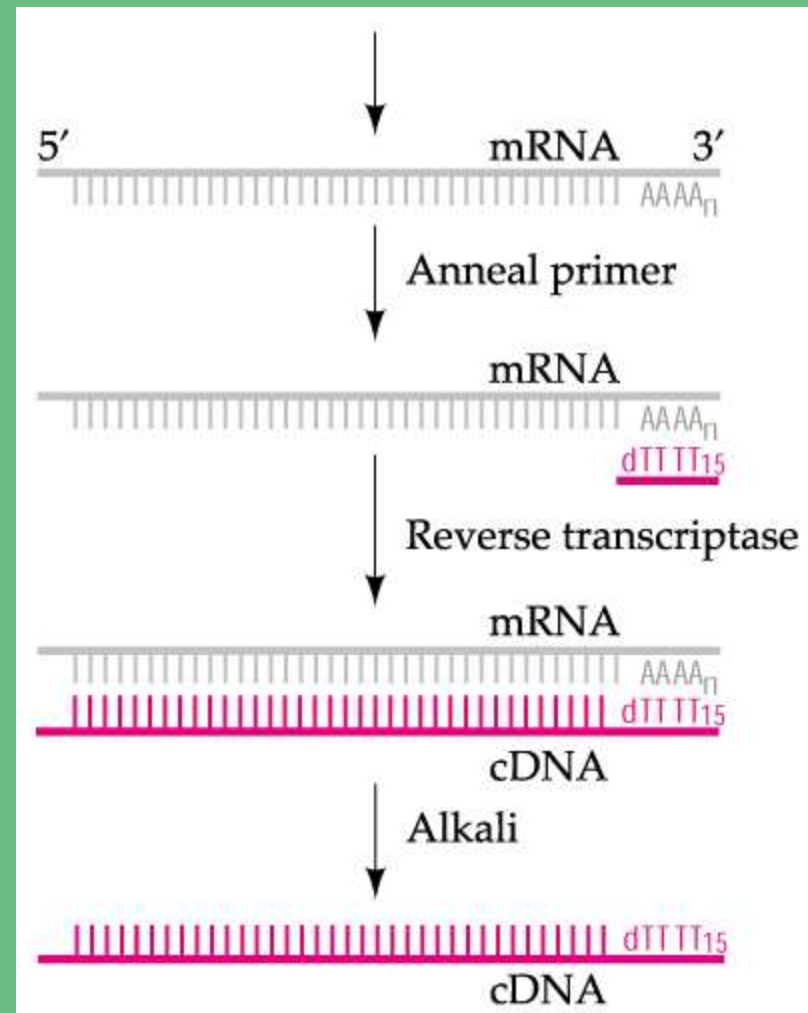
Nucleic Acid Hybridization

- DNA is separated into its two strands
- $C_{\text{RNA}} \gg C_{\text{DNA}}$ the RNA will replace one of the DNA strands in this region



Complementary DNA (cDNA)

- RNA isolated and converted to cDNA
- cDNA isolated
 - raising the pH (*denature helix*)
 - cleaving RNA



Reverse Transcription Polymerase Chain Reaction

- PCR followed by transcription with reverse transcriptase (to convert the RNA to cDNA)
- Expression mapping: determining when and where certain genes are expressed
 - (1) Denaturing at 94°C
 - (2) Annealing at 68°C
 - (3) Extension at 72°C
 - (4) Rinse and Repeat

Options

	Year			
	1	4	7	10
Option				
A	1 sq. mile	4 sq. miles	4 sq. miles	7 square miles
B	7 sq. miles	7 sq. miles	7 sq. miles	7 sq. miles

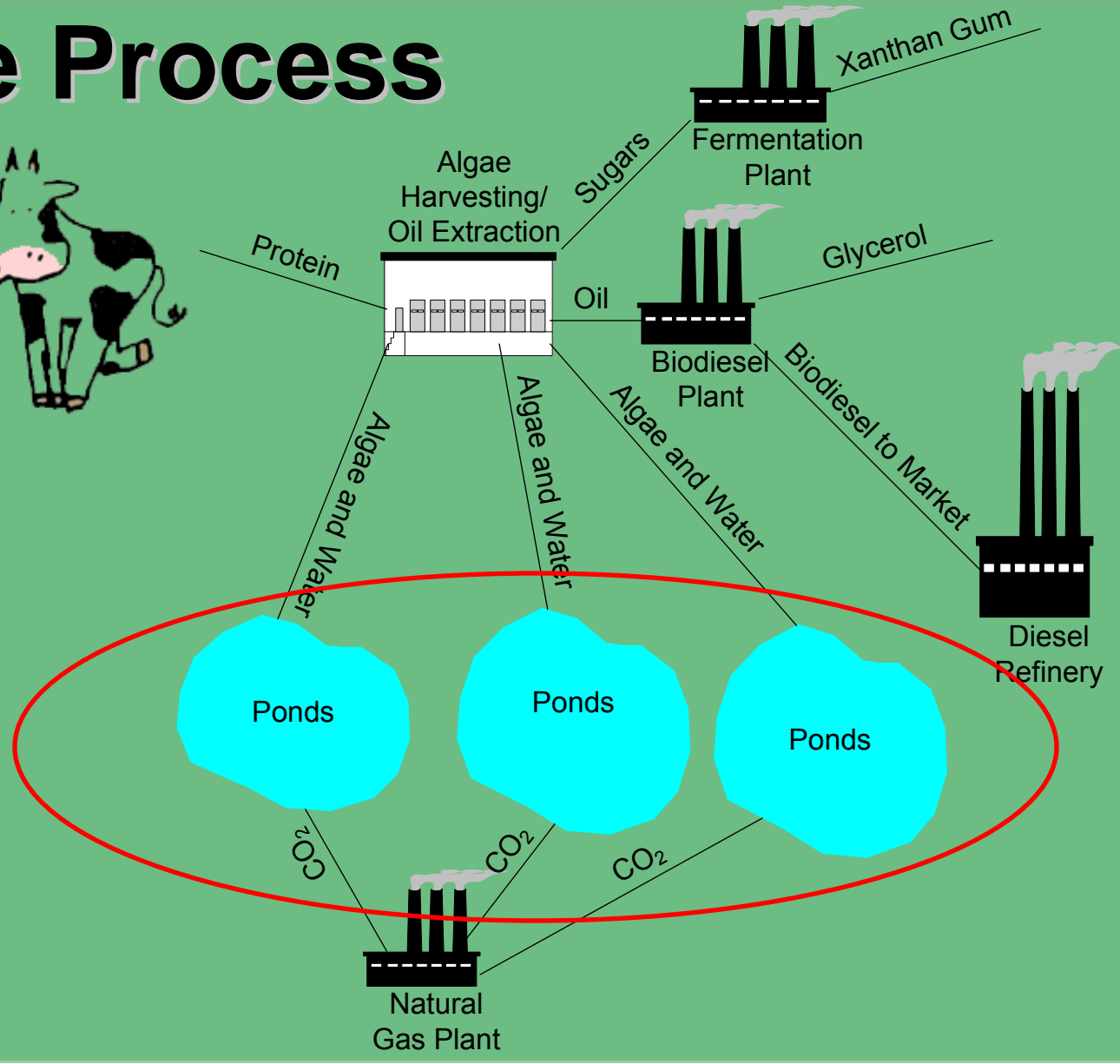
- 15 year production term
- Evaluate profitability based on different starting capital investments
- Expansion vs. One time Build
- Possibility of recycling profits advantage

Options Cont.

Production Rate (million gallons/year)			
Land Usage (sq. miles)	20% Lipid	30% Lipid	40% Lipid
1	1.4	2.1	2.8
4	5.6	8.4	11.2
7	9.8	14.7	19.6

- **Varied Lipid Content to evaluate profitability based on different lipid yields**

The Process

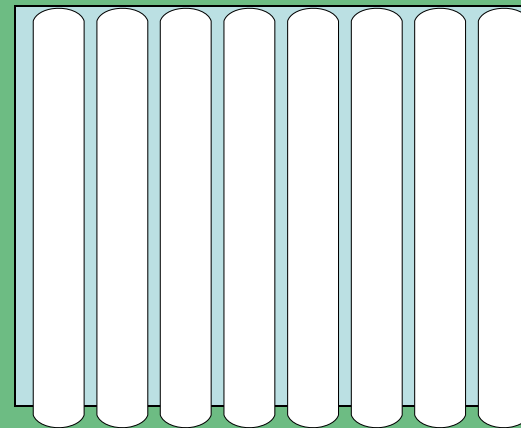
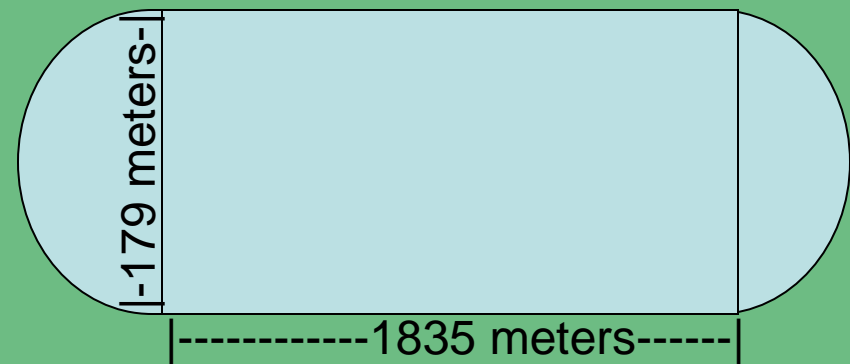


Pond Production Plan

1. 7 day cycle
2. Drain at night
3. Refill pond w/ ~85% old water
4. Pump rest to evaporation pond
5. Add ~15% reclaimed water

Ponds

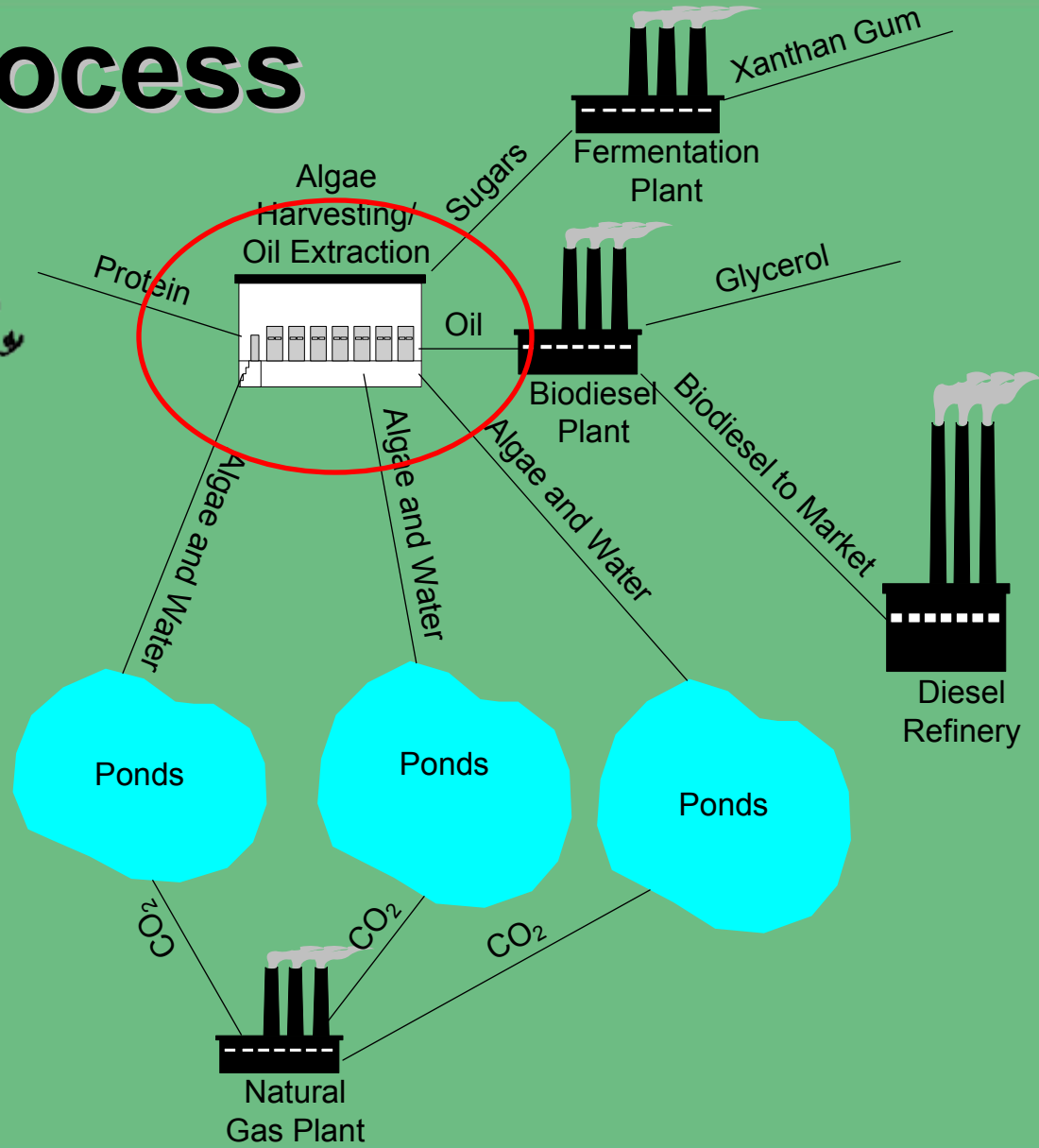
- Length = 1835 m
- Width = 179 m
- Based on 1 mile²
(8 ponds to per mile²)
 - 7 ponds for growth
 - 1 pond for cyclic process



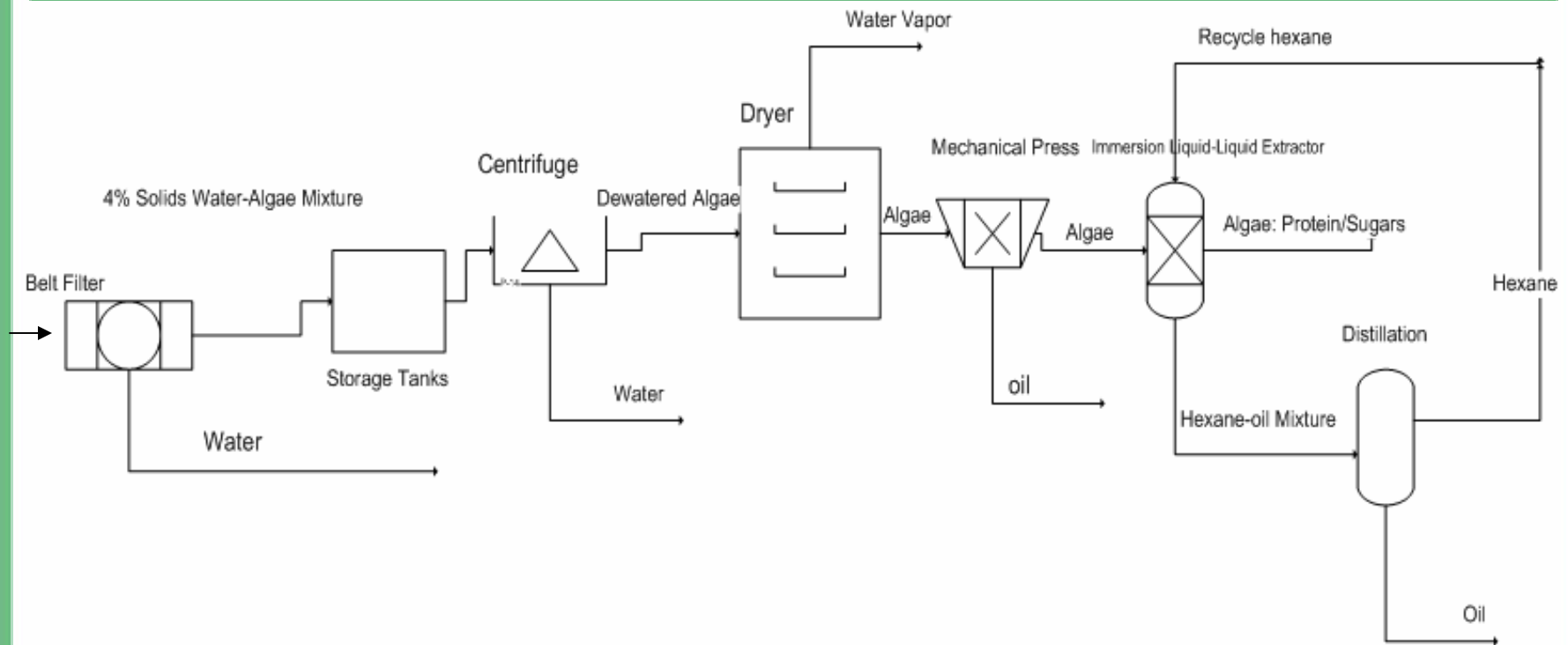
Ponds

- Evaporation Ponds
 - According to NREL
 - West Texas Region 1 cm water/day will evaporate
 - Brine Waste disposed at land fill \$8/yd³
 - Supplement with reclaimed water.
\$0.94/1000 gallons.

The Process

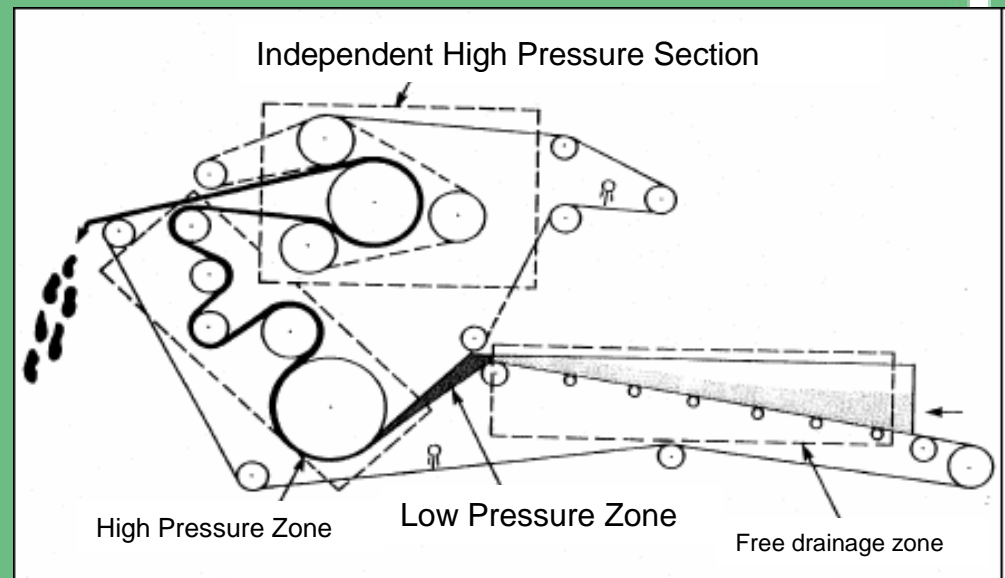


Harvesting / Oil Extraction



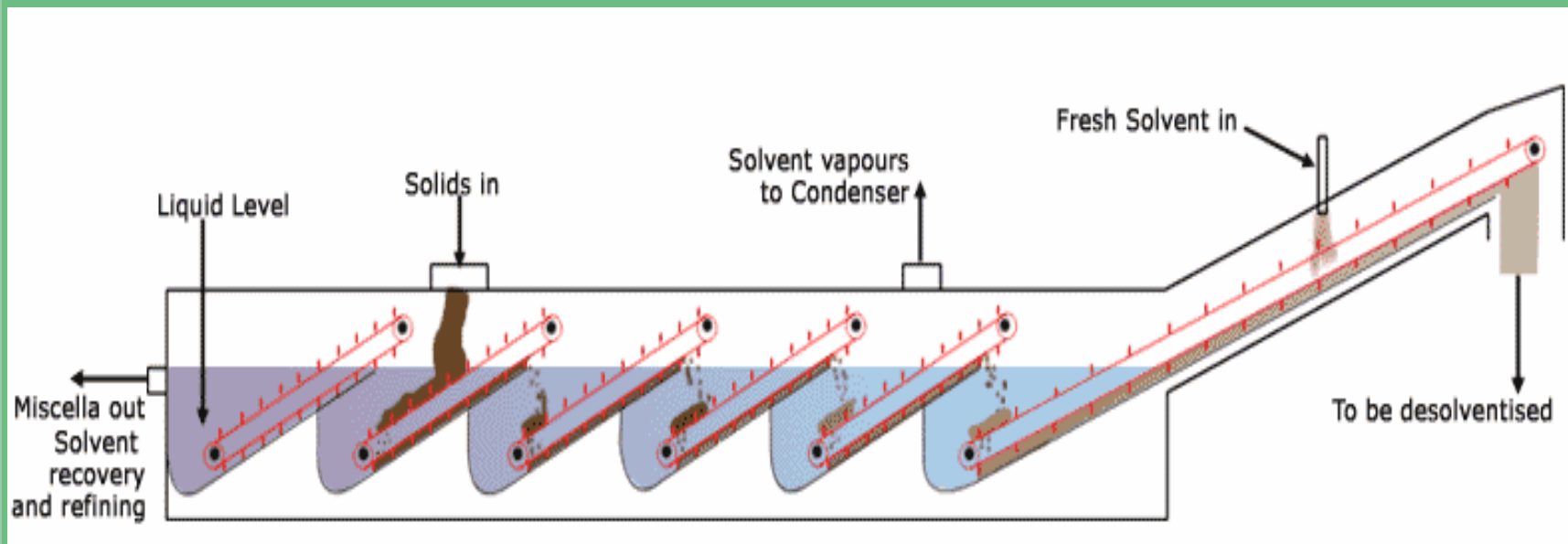
Harvesting

- Empty ponds overnight
- Belt Filtration
(0.04 - 4% solids)
- Centrifuge
(4% - 60% solids)
- Dryer
(60% - 90% solids)



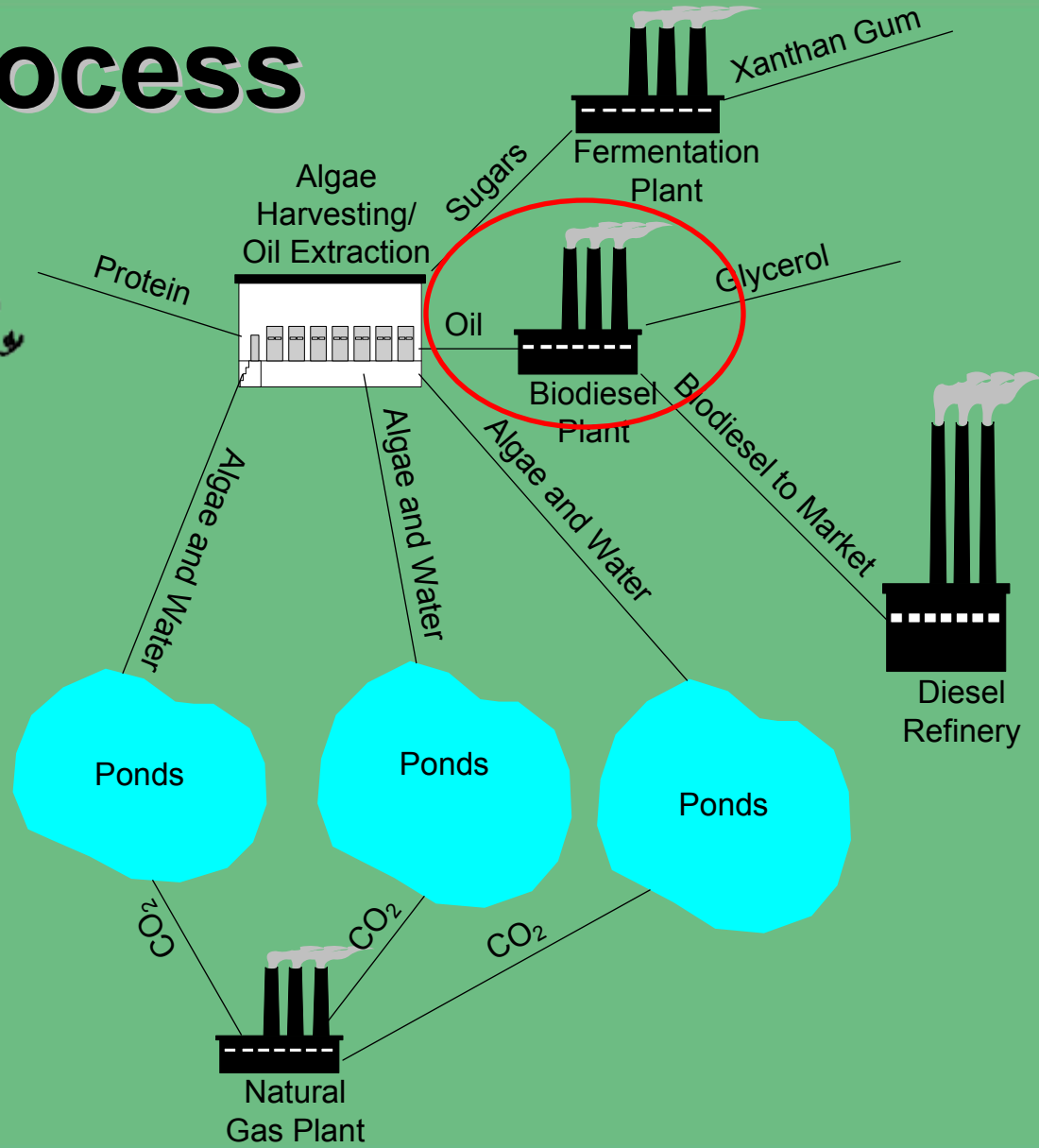
Belt Filtration

Oil Extraction

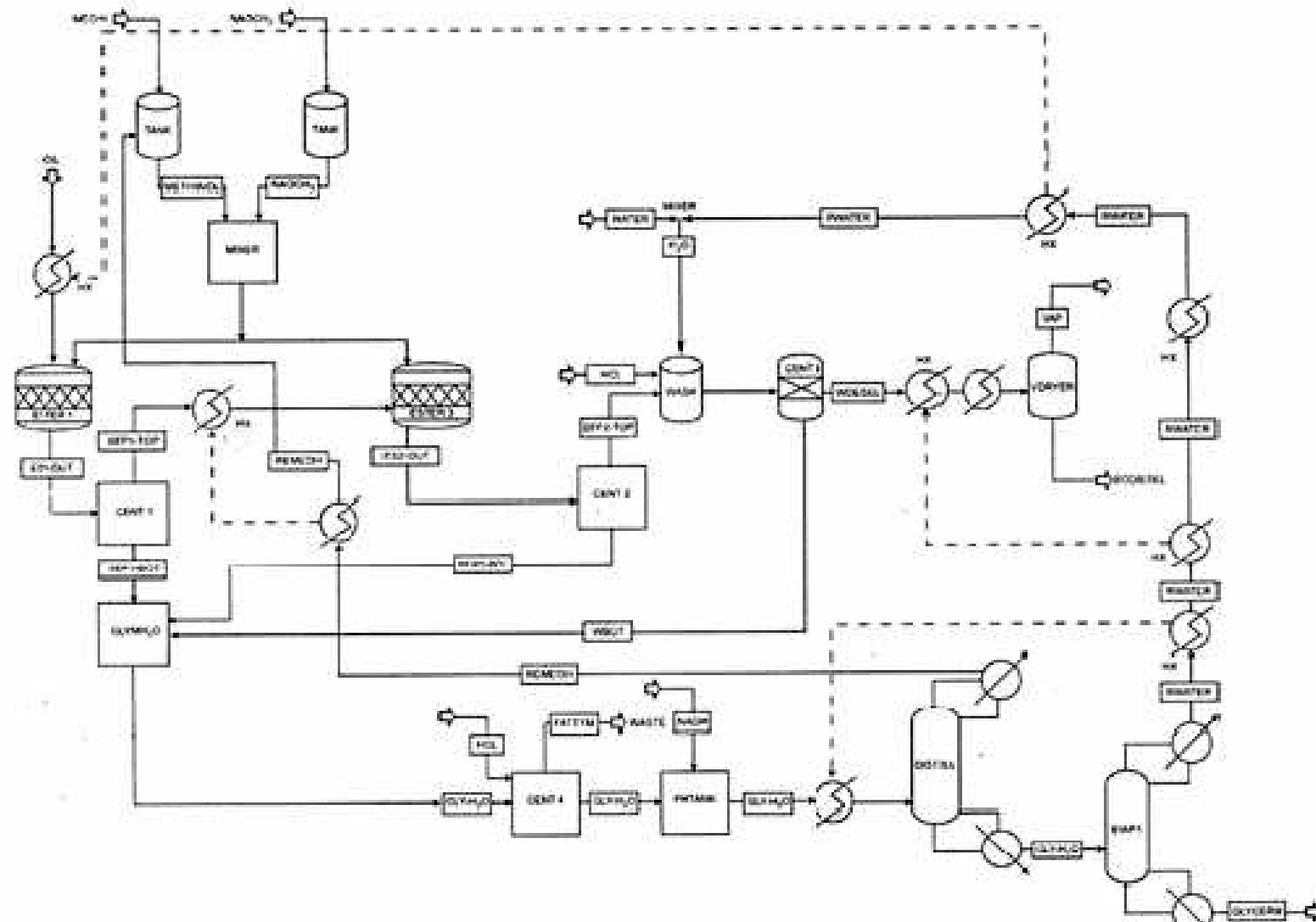


- Mechanical press yields 70% oil
- Immersion extractor w/ hexane yields 95-99% oil

The Process



Transesterification



A process model to estimate biodiesel production costs (Michael J. Haas, Andrew J. McAloon, et al)

Transesterification

- Base Catalyzed Reaction Model
- KOH as catalyst; interchangeable w/ NaOH
- MeOH = cheap, eases separation
- 6:1 ratio of MeOH to triglyceride
 - Highest yield
 - Actual stoichiometry is 3:1
- 90% conversion assume in each reactor
- Yields 99% total conversion

Kinetics

Three reactions that compose transesterification

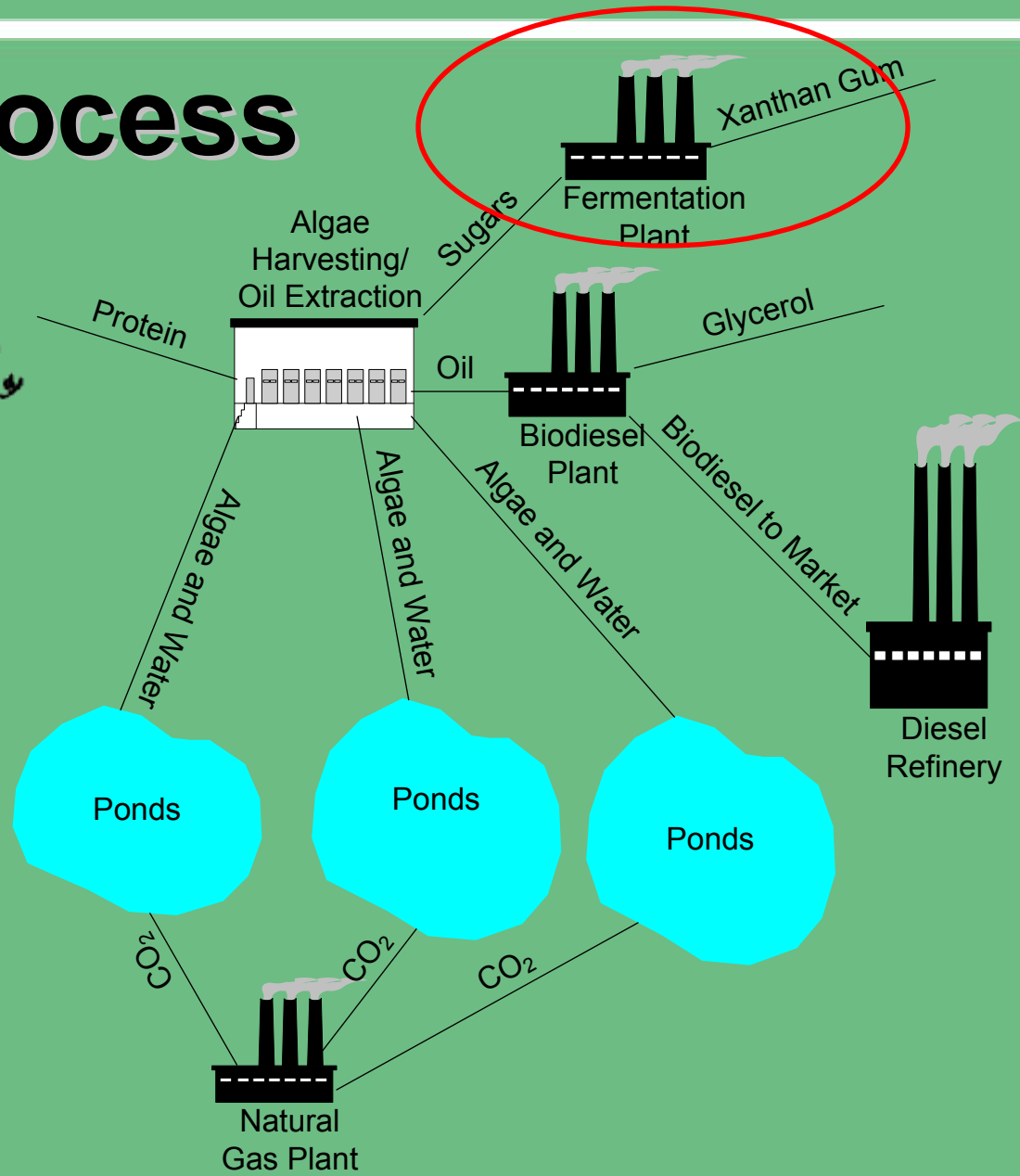
*Tri-triglyceride, Di-diglyceride, Mono-monoglyceride, MeOH-methanol, R_1 -methyl ester 1, R_2 -methyl ester 2, R_3 -methyl ester 3



Mole Balance on Triglyceride Reaction

$$\begin{aligned} 0 = & F_{Tri0} - (F_{Tri0} - \xi_1) \\ & + V[-k_1(F_{Tri0} - \xi_1) / \nu^* (F_{MeOH} - \xi_1 - \xi_2 - \xi_3) / \nu \\ & + k_2(F_{Di0} + \xi_1 - \xi_2) / \nu^* (F_{R1} + \xi_1) / \nu] \end{aligned}$$

The Process



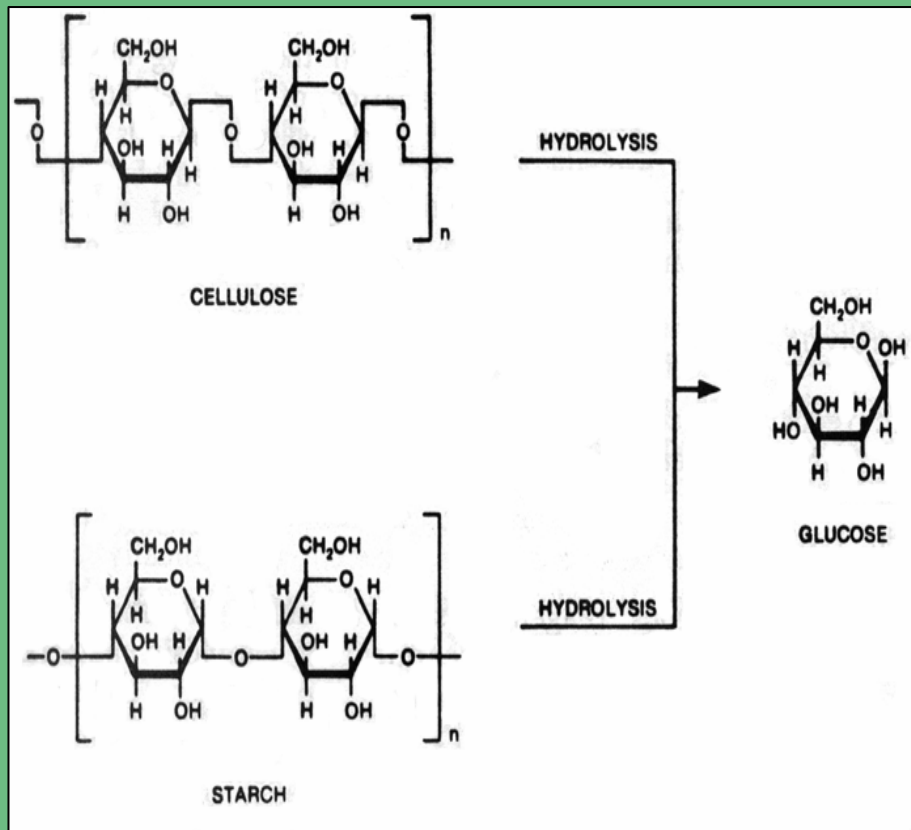
Byproducts

- **What are we going to do with the biomass?**
 - **Pyrolysis of biomass**
 - **Yields bio-oil, charcoal, and flue gas**
 - **Fermentation of carbohydrates / sell protein**
 - **Yields a multitude products**
- **Glycerol produced during transesterification**

Pyrolysis of Biomass

- **Produces Bio-oil (*Don't know its composition*)**
 - Burns at half the heating value of diesel
 - Stability questioned
 - Acidic
- **Questionable Market**
 - \$ 2 MM / yr for 100 tpd facility (*Dynamotive*)
- **Algae contain very small amounts of lignin, an important constituent in the pyrolysis process**

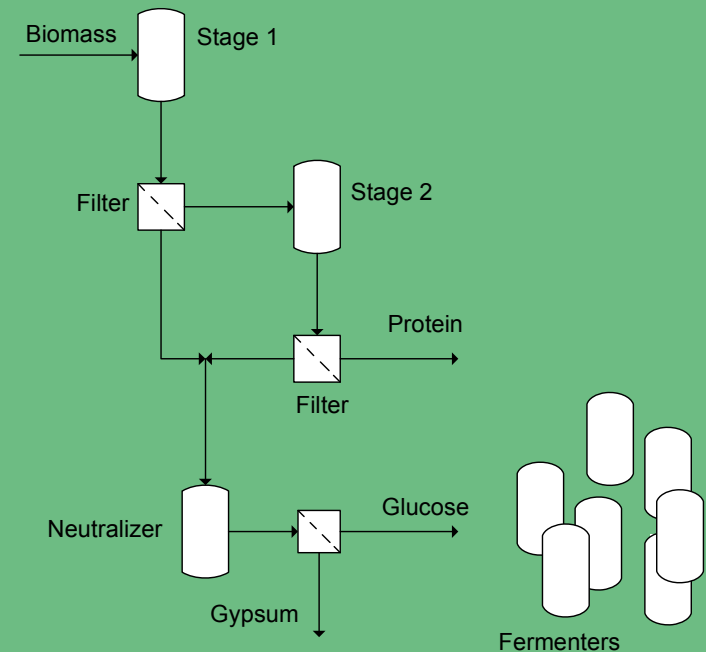
Acid Hydrolysis of Biomass



- Increases glucose yield
- Low concentration acids and high temperatures to process the cellulosic biomass

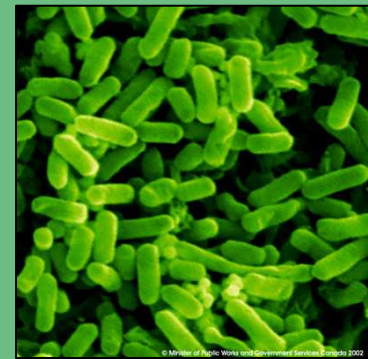
Two Stage Acid Hydrolysis

- **Stage 1:**
 - Targets hemicellulose
 - 0.7% sulfuric acid
 - 190°C
- **Stage 2:**
 - Targets cellulose
 - 0.4% sulfuric acid
 - 215°C

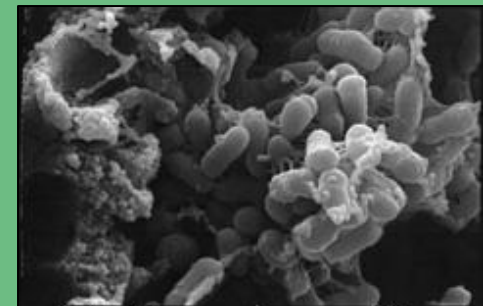


Fermentations

- Succinic acid w/ *Escherichia coli*
 - 59 hour fermentation
 - 0.105 kg glucose/L
 - 54 fermenters at 7 m²



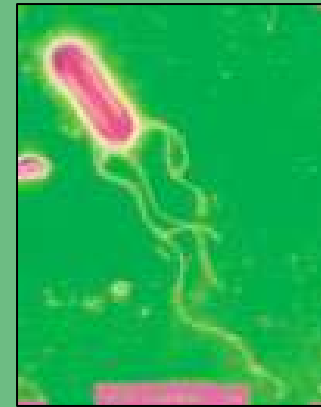
- Propionic acid w/ *Propionibacterium freudenreichii* ssp. *Shermanii*
 - 148 hour fermentation
 - 0.02 kg glucose/L
 - 115 fermenters at 7 m²



$$\frac{(kg_glucose/day)\left(\frac{hrs_fermented}{24hrs}\right)}{1000(kg_glucose/L)(m^3/fermenter)} = \#_fermenters$$

Fermentations Cont.

- 2,3-Butanediol w/ *Aerobacter aerogenes*
 - 0.195 kg glucose/L
 - 35 fermenters at 7 mi²
 - 390 tons at \$1.98/kg

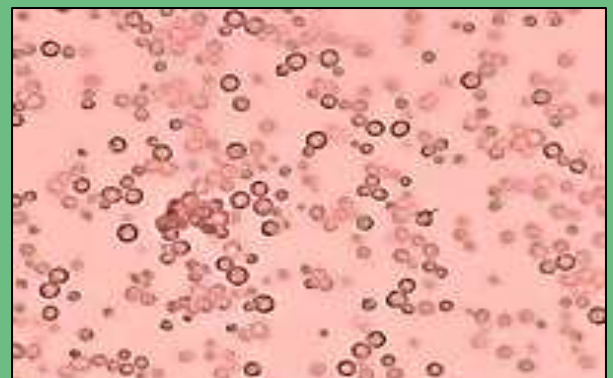
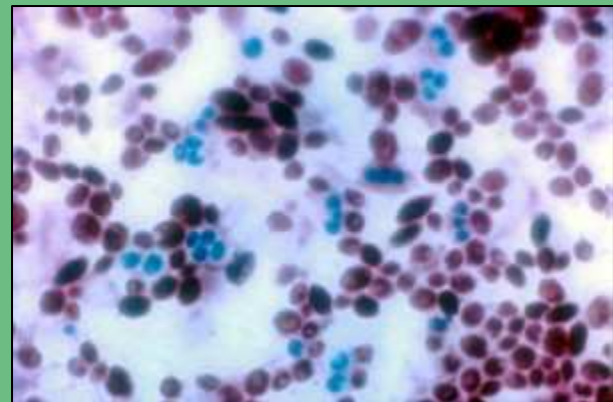


- Butyric acid w/ *Clostridium butyricum*
 - 48 hour fermentation
 - 0.016 kg glucose/L
 - 145 fermenters at 7 mi²



Fermentation Cont.

- Ethanol w/ *Saccharomyces cerevisiae*
 - 18 - 94 hour fermentation
 - 0.05 - 0.2 kg glucose/L
 - 5.1 - 91.8 g ethanol/L
 - 16 fermenters at 7 mi²
 - Selling price \$1.91/gallon
 - 30,000 gallon/batch
 - \$13 M/yr (revenue)



Hyaluronic Acid

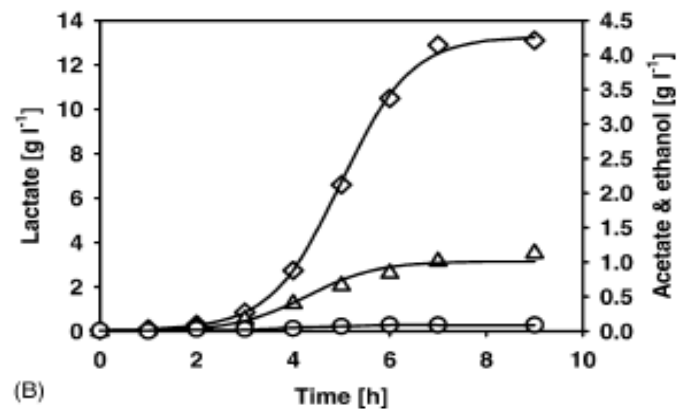
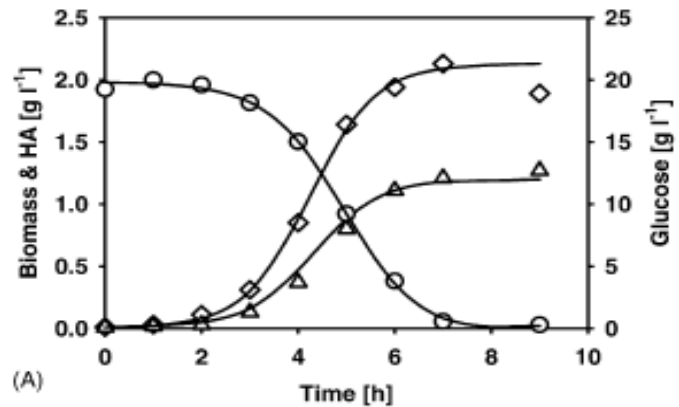


Fig. 3. Growth and product profiles for cells grown in an aerobic batch culture containing 20 g l⁻¹ glucose. The main metabolic products were lactic acid and acetic acid. HA was produced in a growth-associated manner until the glucose was depleted.

- Low temperatures (28°C) and high glucose concentration (40 g/L)
- Hyaluronic acid \$100,000 / kg
- Small market requires extreme purity

*Aerobic cultivation of *Streptococcus zooepidemicus* and the role of NADH oxidase
Barrie Fong Chong, Lars K. Nielsen

Xanthan Gum

Xanthan Gum: \$ 11 / kg
 Market volume: 40 – 50 M tons/yr

Multiple industries:
Food: Beverages, Dairy, Sauces, Meats, Frozen Deserts, Bakery

Consumer / Industrial: Cleaners, Oral Care, Paints, Cosmetics, Pharmaceutical, Printing

Oil: Mud's / Drilling Fluids

Paper:

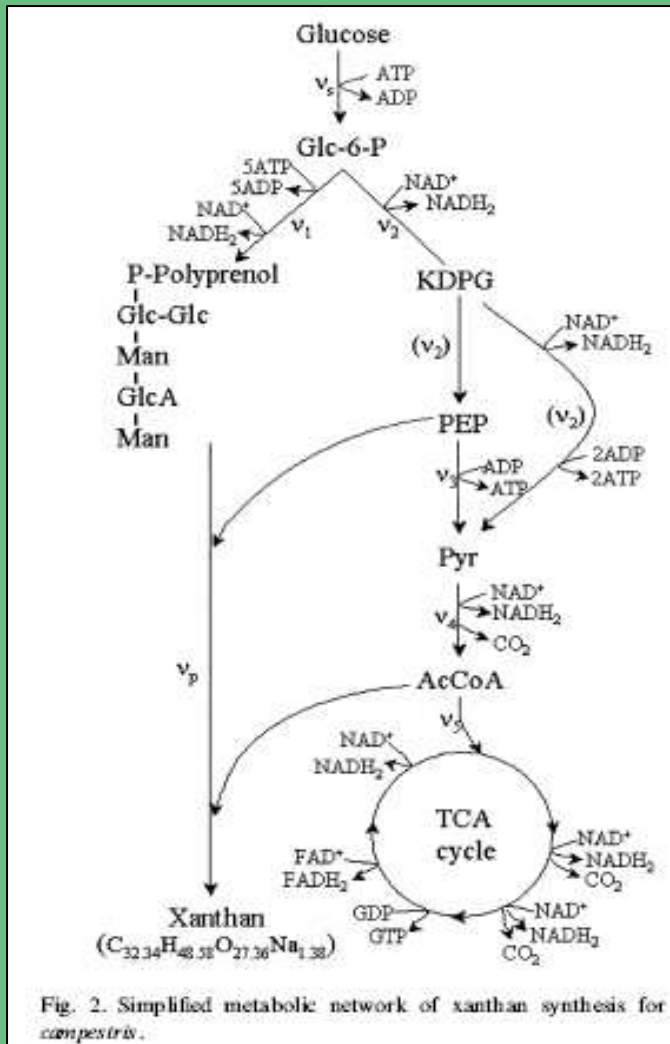


Fig. 2. Simplified metabolic network of xanthan synthesis for *campestrii*.

Characterization of xanthan gum biosynthesis in a centrifugal, packed-bed reactor using metabolic flux analysis
 Chia-Hua Hsu, Y. Martin Lo

<http://www.apsnet.org/education/IntroPlantPath/Topics/plantdisease/images/fig07.jpg>

Traditional Xanthan Fermentation

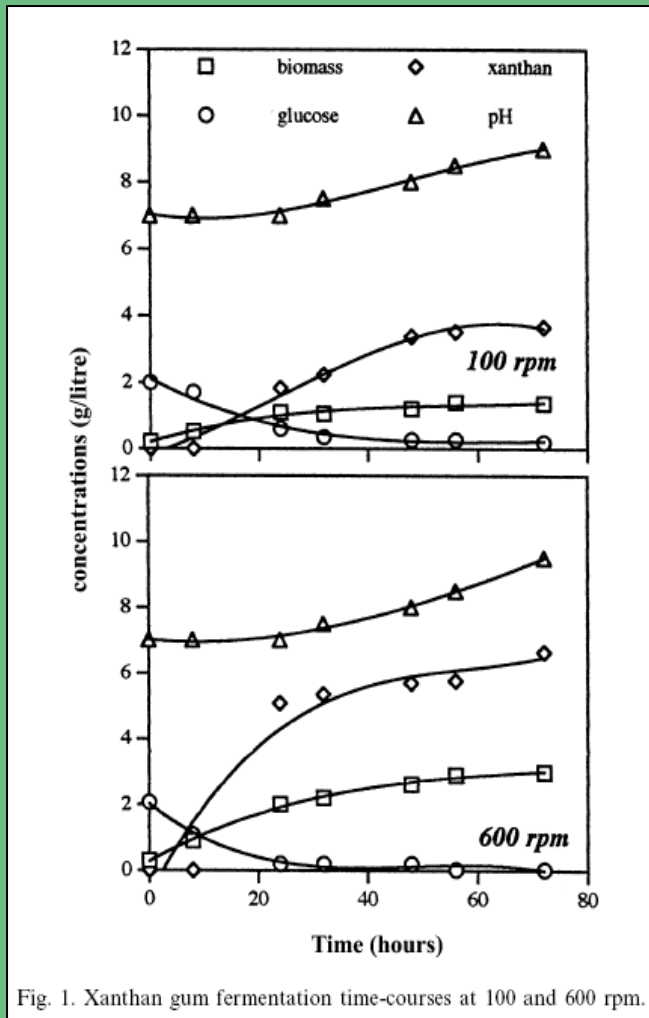


Fig. 1. Xanthan gum fermentation time-courses at 100 and 600 rpm.

- Production dependent upon mixing speed
- Aeration is an issue
- 72 hour fermentation
- 0.05 kg glucose/L
- 77 fermenters at 7 mi²

Xanthan production by *Xanthomonas campestris* in batch cultures

M. Papagianni a, S.K. Psomas a, L. Batsilas a, S.V. Paras a, D.A. Kyriakidis b, M. Liakopoulou-Kyriakides

CPBR Xanthan Fermentation

- 24 hour fermentation
- 0.05 kg glucose/L
- 26 fermenters at 7 m²
- Repeatable fermentations once cells adsorbed
- Cell-free broth

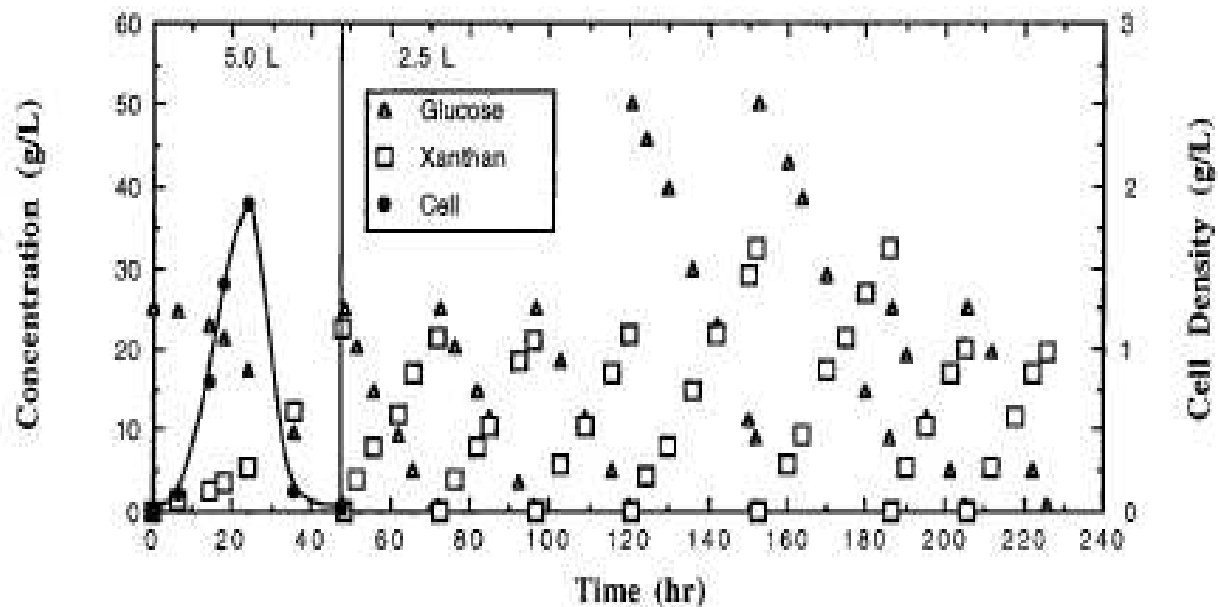
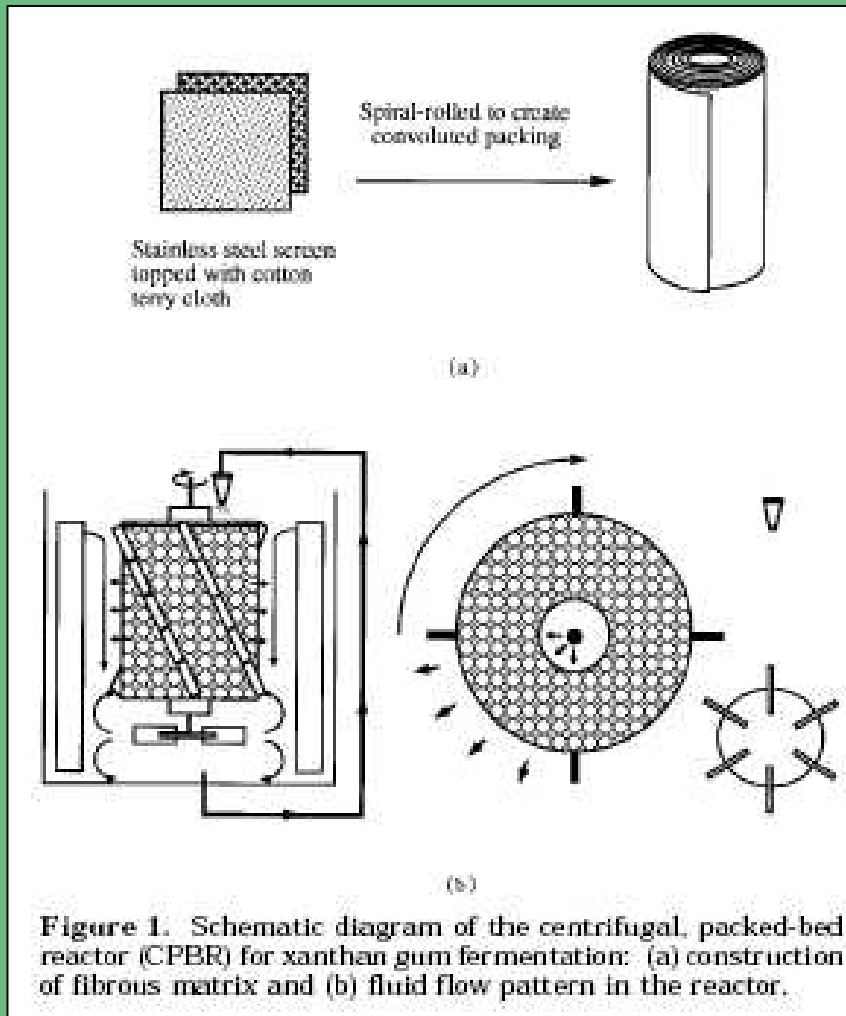


Figure 6. Fermentation time course data for repeated batch xanthan fermentations with CPBR at 350 rpm rotational speed for the fibrous bed under gas continuous condition (CPBR-GC).

Centrifugal Packed Bed Reactor



- Improves the energy-intensive, low-yield process due to limited aeration
- Produces cell-free broth with elevated productivity

Xanthan Gum Fermentation by *Xanthomonas campestris*
Immobilized in a Novel Centrifugal Fibrous-Bed Bioreactor
Shang-Tian Yang, Yang-Ming Lo, and David B. Min

http://www.vikingpump.com/products/rotary_lobe_pumps/images/LobePumpLarge.gif

Xanthan Production

- 4.4 tons xanthan gum / batch

Ultrafiltration of xanthan gum fermentation broth

221

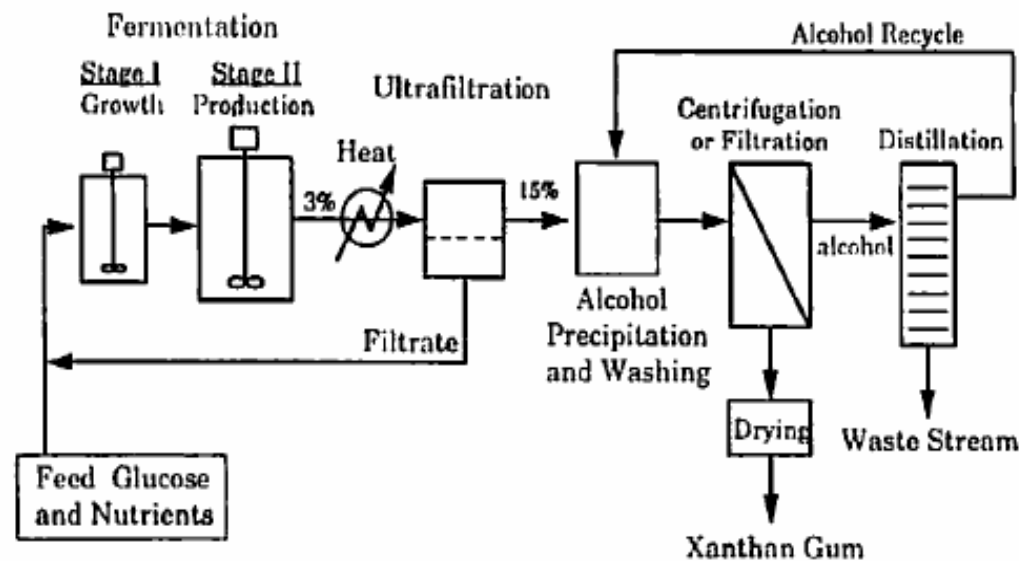


Fig. 1. A process flowsheet for xanthan gum production from fermentation, ultrafiltration and alcohol precipitation.

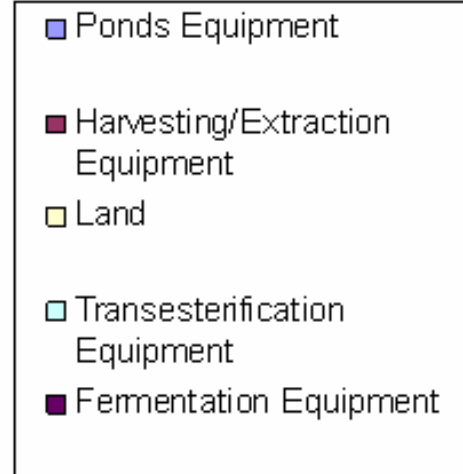
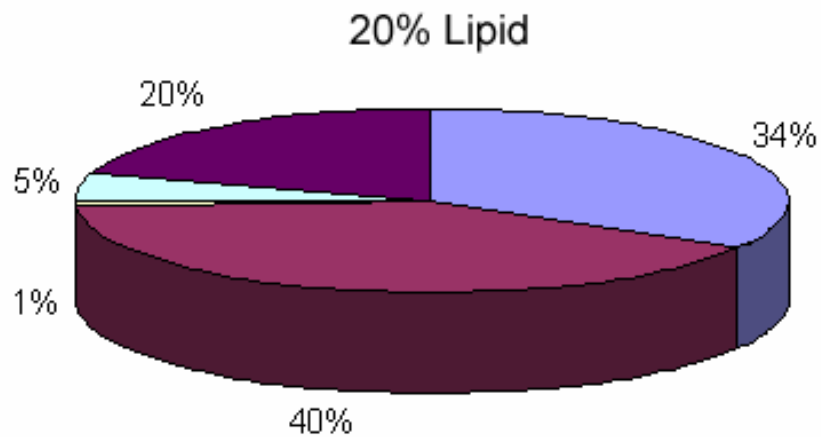
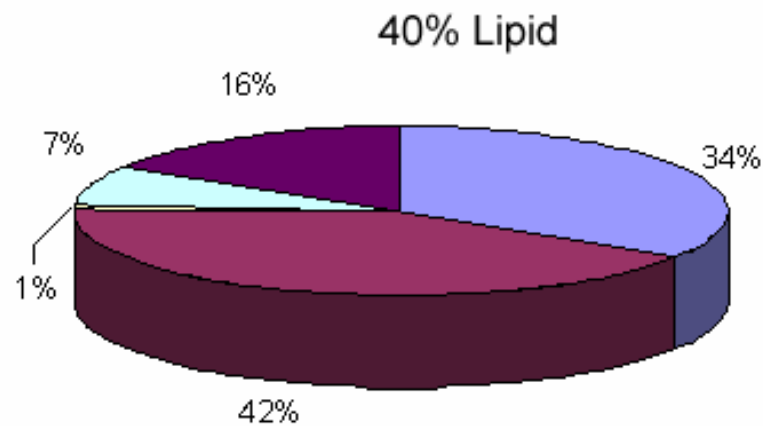
What to do with the glycerol?

- Foods and Beverages
 - Soft Drinks, Candies, Meat & Cheese Casings
- Drugs
 - Capsules, Suppositories, Lozenges, Gargles
- Cosmetics / Toiletries
 - Emollient (softening agent)
 - Moisturizer
- Tobacco
- Paper / Printing
- Textiles
 - Lubricant

Sell at \$0.15/lb



Cost Breakdown



Economics

Option	Lipid Content	TCI	FCI	NPW	ROI
A	20%	574,000,000	488,000,000	-304,000,000	-53%
	30%	569,000,000	484,000,000	-329,000,000	-58%
	40%	563,000,000	479,000,000	-336,000,000	-60%
B	20%	364,000,000	309,000,000	113,000,000	31%
	30%	362,000,000	308,000,000	47,000,000	13%
	40%	359,000,000	306,000,000	24,000,000	7%

- Biodiesel: \$2.00/gallon
- Payout Time
 - Option B : 5 - 7

Pricing

Cost B20 =

$$\begin{aligned} & \% \text{Diesel} * \text{Production} * \text{Cost} \\ & + \% \text{BioDiesel} * \text{Production} * (\text{Cost} - \alpha) \end{aligned}$$

Δ Profit =

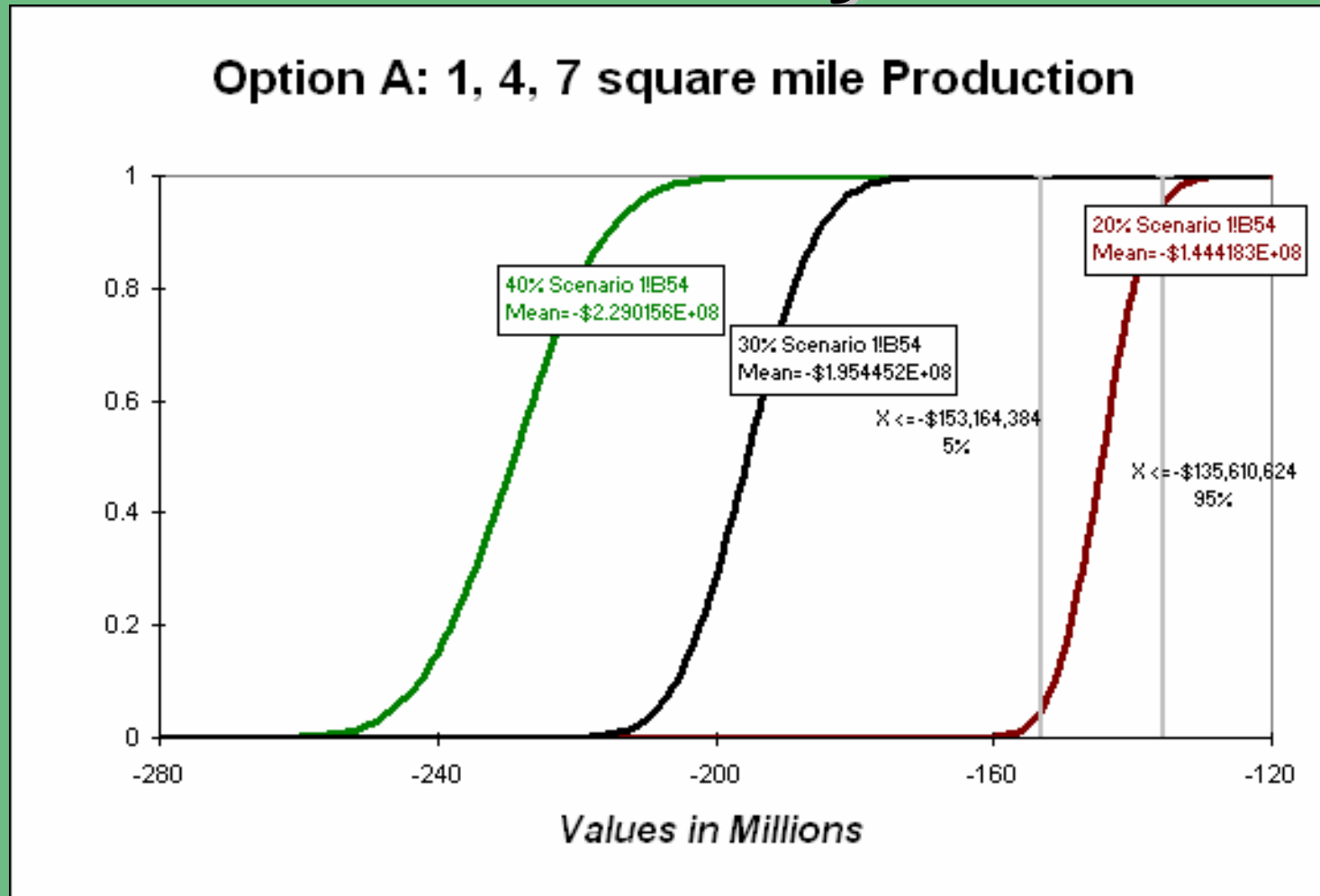
$$\begin{aligned} & \text{Production} * (\text{Selling Price} - \text{Cost B20}) \\ & - \text{Production} * (\text{Selling Price} - \text{Cost}) \end{aligned}$$

Solved for α such that their equipment costs are paid off in one year ($\alpha = .05$)

Sell to biodiesel to refinery for B20 (20% Biodiesel) blend
\$0.72/gallon

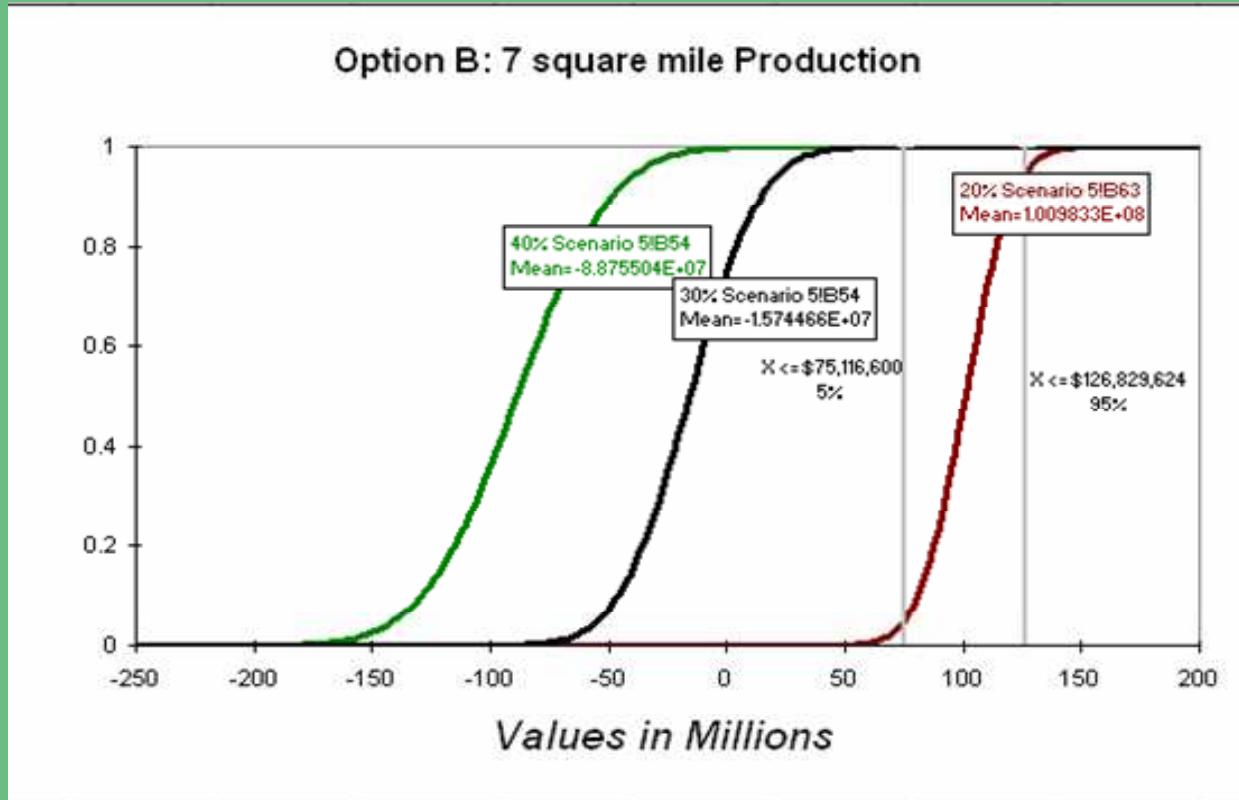
Risk Analysis

Option A: 1, 4, 7 square mile Production



- Risk Curves, and Minimax
- Explain high TCI and low ROI

Risk



- The risk is extremely large since everything relies on production of algae.
 - Cold weather
 - Alternative algae take over
 - High Cost of Making Ponds and Harvesting Algae
 - Overall high TCI and long payout time
- Varied Selling price of Biodiesel

Price Effects

		Net Present Worth (15 years)			
Selling Price of Biodiesel		\$0.72	\$1.50	\$2	\$3.14
Option A	20% Lipid	-\$356,000,000	-\$324,000,000	-\$304,000,000	-\$257,220,000
	30% Lipid	-\$407,000,000	-\$359,000,000	-\$329,000,000	-\$258,950,000
	40% Lipid	-\$440,000,000	-\$377,000,000	-\$336,000,000	-\$243,200,000
Option B	20% Lipid	\$13,000,000	74,000,000	\$113,000,000	\$202,800,000
	30% Lipid	-\$104,000,000	-12,000,000	\$47,000,000	\$181,100,000
	40% Lipid	-\$177,000,000	-55,000,000	\$24,000,000	\$203,200,000

- As biodiesel price increases NPW goes up
- \$3.14/gallon biodiesel becomes more profitable than byproduct fermentation
- \$9.50/gallon Option A becomes Profitable

Conclusion

- Algae
 - Good source for oil
 - High Cost Harvesting
 - High TCI
 - Profitable but risky
 - Further research is necessary to make the process more economically feasible

Recommendations

- Evaluate algae types with onsite experimentation
- Explore the cost of bioreactors with the rising diesel prices
- Explore more economical ways of harvesting
- Utilize Tax Incentives



Questions?